

**FORMING A COMPLETE RECORD OF THE PROCEEDINGS OF ALL PUBLIC COMPANIES.**

[PRICE 6D.]

James J. O'Brien, Waltham, Mass., 1901.



## LAW INTELLIGENCE.

## SOME OF THE ADVANTAGES OF BECOMING A DIRECTOR.

MIDLAND CIRCUS—WARWICK, MARCH 26.

**BOULTON & SHAFER.**—The plaintiff had been some time ago a respectable surgeon in good practice at Leamington. In 1835 he had the misfortune to become a member and a director of the Leamington Joint-Stock Banking Company, then formed with a nominal capital of 200,000*l.*, divided into 10,000 shares of 20*l.* per share. In 1836 this bank was in difficulties, and the directors and managers borrowed 50,000*l.*, for 40,000*l.*, of which the plaintiff gave his bond jointly with two other persons, and the banking company gave at the same time a promissory note for the 50,000*l.*. When that note became due the bank could not pay it, having stopped payment, and the London and Westminster Bank, in London, to which the note and bond had come, sued Mr. Boulton, who defended the action unassisted by the company, and recovered a verdict and judgment against him, and his effects were sold under execution. For the sum of 3947*l.*, including the costs, thus recovered from him, and also for damages, he brought his action against the public officer of the said bank. The defence was various, but chiefly that Mr. Boulton owed the bank, for calls on a large number of shares held by him, and that he not only would not pay these calls on his own shares, but encouraged other shareholders not to pay their calls. The defendants also charged the plaintiff with fraud in transferring his shares at a time when he was chairman of the board of directors.

Lord Abinger told the jury that he was of opinion the transfer of the shares was not legal, though there was no fraud, and that this being an action for damages, the debt due by calls on the shares could not be set off, except against so much of the action as sought to recover a balance due to the plaintiff on his banking account.

A great number of issues having been raised in the pleadings, many witnesses were examined on both sides, and various points of law were strongly contested. The jury found a verdict for the plaintiff for the sum which he had to pay by force of the execution issued against him by the London and Westminster Bank, but they did not think him entitled to any damages.

## SPECIFICATIONS OF RECENT PATENTS.

[From the *Inventors' Advocate*.]

**Walter Richardson, gent.,** Regent street, and George Mott Brithwaite, gent., Manor house, Chelsea, for improvements in tinning metals (communicated by a foreigner). March 17.—This invention relates to a mode of combining nickel and iron with tin, in order to improve the tinning of metal surfaces. The proportions of nickel and iron, mixed with the tin, in order to produce the best tinning, are—ten ounces of the best nickel, and seven ounces of about iron, to ten pounds of tin.

The temperature at which nickel is fusible being higher than that required to bring tin into a state of fusion, it is necessary to prevent the tin as it melts from evaporating (as it is essential that the two metals be put into the same crucible); this object is attained by adding to ten pounds of the composition one ounce of borax and three ounces of powdered glass; the heat soon causes the borax to bubble up, which augments its volume, and causes it to unite with the melted glass without mixing with the metals. The lightness of these two substances in comparison with the metals, causes them to rise to the surface, where they form a crust, which prevents the action of the air on the metals, the fusion of which proceeds under the influence of a concentrated heat. The fusion is completed in about half an hour, when the composition is run off through a hole in the crust. In tinning metals with this composition the workman proceeds in the ordinary manner.

**Paul Hennele, solicitor, of Paris,** and now of Clement's-lane, London, for improvements in the construction of governors or regulators, applicable to steam engines and to other engines used for obtaining motive power. March 10.—Claim first.—The construction and arrangement of an inflated reservoir cylinder or receiver, supplied with atmospheric air, gas, water, or other liquid, by means of the moving power, the supply of which it is designed to regulate.

Claim second.—The combination of such inflated reservoir or other governor, with a valve or damper in the furnace flue, by which the intensity of the fire is regulated according to the speed of the engine or the supply of steam, which damper may be variously constructed and placed in the furnace flue or chimney, or in front of the furnace, according to the convenience or locality of the engine to be governed.

Claim third.—The peculiar construction of the moveable shuttle, or water-gate for hydraulic wheels, together with such modifications of the same as may be required in peculiar localities, by which the operation of this or other regulators is rendered available to hydraulic wheels.

The regulator is a frame containing a bellows or elastic reservoir, connected with two similar pieces of bellows underneath it. They are formed of leather, and are secured to horizontal plates or bases by iron straps. The middle plate which separates the reservoir or upper bellows from the lower bellows is fixed, but the two other plates, viz., the top plate of the reservoir, and the plate that separates the two bellows, are moveable. A rod or central shaft rises from the top plate, and to it is fastened a chain, which, passing over a pulley, is attached to a sector provided with a counter weight. This sector is connected with the throttle valve of the engine, so that any vibration of the sector opens or shuts the throttle valve, and governs the distribution of steam. A second chain or cord, attached to the central shaft, is carried over other pulleys, and transmits to a vertical shaft and register placed in the flue of the furnace a rotary motion, opening or shutting the flue of the furnace according to the elevation or depression of the reservoir, and thereby regulating the furnace in proportion to the amount of steam consumed.

To the base of the machine are firmly attached two supports, which receive the journals of a shaft provided with a double crank; and a fast and loose pulley, fastened on this shaft, carry a strap from any convenient part of the engine or first mover to be regulated. The cranks on the shaft are provided with connecting rods, which convey a reciprocating motion to two perpendicular rods guided in the ears of the fixed plate, and fixed by pins in the ears of the lower plate, i. e., the plate that separates the lower bellows; so that if any impulse be given to the horizontal or crank shaft, the connecting rods transmit a reciprocating movement to the perpendicular rods, which give an alternate up and down movement to the lower plate.

When the moveable plate is lifted up the air enters the bottom bellows through a valve, while by the same movement the air which was contained in the middle bellows will have been forced forward through a valve into the reservoir, and on the other hand when the lower plate descends, the air which has been received into the interior of the bottom bellows, will be forced forward through a leather pipe into the reservoir, while a fresh supply of air will be received into the middle bellows through an horizontal channel in the lower plate. By this means the air is alternately forced forward by each of the bellows to the reservoir.

Thus the principal function of the regulator is to accumulate in a common reservoir the air of two or more bellows, in order that the top plate lifted by the pressure of the air may communicate (according to the variation of the speed of the engine to be regulated) an ascending or descending motion to the central shaft, which communicates by means of chains, &c., with the throttle valve of the steam engine, or the shuttle or water gate of the hydraulic wheel.

The escape of air from the reservoir is regulated by a valve safely adjusted. The form of the water-gate is either rectangular, curved, or cylindrical.

**GRAND NORTH OF ENGLAND RAILWAY.**—This railway (which carries forward the communication from York to Darlington, and completes the chain from the metropolis to the county of Durham) was opened for the conveyance of passengers on Wednesday last, the 31st ult. The 5th of April being fixed by the post-office authorities for the acceleration of the mails from London to the north, and the Grand North of England line being included in the mail contracts, all places as far north as Darlington, and indeed beyond, will share in the benefit of the increased rapidity of post-office communication.

**MARYPORT AND CARBIDE RAILWAY.**—We are informed that the additional part of this railway, leading from the present terminus to the Oughthorpe coal field to Ayrshire, will be opened on Easter Monday. This will afford many advantages and facilities to travellers and to the people of the district; and the company intend shortly afterwards, we learn, to start an omnibus from the railroad at Ayrshire to Wighton and Carlisle, for the speedy conveyance of passengers and goods. These things show that the genuine spirit of enterprise now animates the promoters of this undertaking, which we expect to see go forward to completion with a rapidity and success that will surprise its opponents, but afford the highest delight and gratification to its numerous friends and the public at large. —*Whitaker's Herald*.

**LENGTH OF DAYS.**—At Berlin and London, the longest day has six hours and a half. At Stockholm and Upsal, the longest has eight hours and a half. At Hamburg, Danzig, and Berlin, the longest day has seven hours, and the shortest seven. At St. Petersburg and Tobolsk, the longest has six hours, and the shortest five hours. At Tromsø in Finland, the longest day has twenty-one hours and a half, and the shortest two and a half. At Wadswick, in Norway, the day lasts from the 21st of May to the 23d of July, without interruption; and in Spitzbergen, the longest lasts three months and a half.

## ON THE CONGELATION OF ROCKS.

A very interesting discussion took place on some appearances connected with the congelation of rocks, at the last monthly meeting of the Manchester Geological Society. Dr. BLACK having recapitulated the principal points in a paper submitted by him at a previous meeting, produced an explanatory diagram, and then noticed the disintegration of the soft sandstone into laminae, especially that species which was cut up into sandstone slabs, which was taken from the upper surface, split up by the workmen, and cut into proper shape for roofing; but independently of this, he said, there seemed to be a change which took place in the surface of the rock, after having been raised or lifted up from the original bed, as if it had been subjected to a severe degree of congelation, intermitted with thawing, till the superficial beds of these strata were loosened one from another, and subsequent thawing made them settle more or less, and very likely a succeeding freezing would still further enable the water to percolate through the interstices, and the stratifications were found loosened and disintegrated, but still lying parallel to each other. Sometimes angular pieces of four, six, and even up to twelve inches, were found turned about from their vertical axis, and all exhibiting angular projections. Now, he attributed the cause of these appearances to have been a great body of water moving, at some distant period, over the surface of this rock with considerable force, but not with sufficient velocity to have torn the beds in situ, as they were found in several directions. These were altogether of the same nature as the subjacent strata, although at first sight they might be supposed to have been rocks brought from a distance; but, upon examination, he found these disintegrated portions to belong to the mother rock below. This body of water had brought with it more or less detritus from other rocks, such as boulders with the angles rounded off, showing that they had been subjected to the action of water. These were sometimes found in the surface above, even from schistose rocks, mixed with those of more recent growth, such as some of the sandstone rocks, the edges being rubbed off. He had also found little stratifications of coaly matter, and in one or two instances transported coal fossils in the diluvial, above the disintegrated portions of rock. The depth of the diluvial deposit above, varied from six inches to six feet; and when the rock approached very near the encampment, a culminating point of the strata, it would even be bare. The supposition he had suggested would account for many of the diluvial deposits found in mountainous districts, independently of the proof they seemed to him to furnish of congelation. After congelation, the surface of the rock had been covered by the deposit contained in the great body of water moving over it, and tearing up the superficial part of these loose beds, and depositing its own load as it came over. The sand was not in the line of the disintegrated bodies, but horizontal; and the boulders were lying upon it. The disintegrated pieces accumulated at the declivities by the action of gravity. He thought the facts adduced altogether confirmatory of the supposition of this body of water having moved over the whole surface of the rock, and produced these different effects, after the rock had been in the state of congelation of which he spoke. One diagram in Mr. De la Beche's work showed such an appearance affecting the slaty rocks in Cornwall; but he did not there say anything of congelation, but attributed it entirely to the surface being covered with water. In conclusion, Dr. Black read a passage, referring to similar appearances, from Mr. McLaughlin's *Notes to accompany a Geological Map of the Forest of Dean Coal-Field*.

Mr. BOWMAN said, that he thought the appearance so ably described by Dr. Black was not difficult to account for. It was well known to zoologists, that all, or very nearly all, sedimentary rocks possessed distinct lines of stratification; but that these were frequently not apparent, especially in the older formations, till the surface had been for some time laid bare and exposed to atmospheric influences. The consolidation was so great that the divisional planes were not perceptible till the rain, the frost, and the disintegrating agents in the air, had penetrated within and between the beds, and, by oxidising the iron in the stone, had given them a brown rusty hue. He thought, therefore, that large blocks or portions of the rock in question had, in the first place, been thrown out of their natural and conformable position, probably by the erosive action of strong currents; and that afterwards, when left dry, the rain, frost, &c., had acted on these blocks, and separated them into many parallel divisional planes or layers. These were no other than the lines of original bedding, which appeared to diverge upwards in series from the general direction of the stratification, only because the blocks had been disturbed while entire, before they were separated; and the latter retained their parallelism, because they were subsequently buried up under a mass of diluvium, which protected them from further derangement. As this was a very curious subject, and intimately connected with the external characters of rocks, Mr. Bowman shortly alluded to the natural separation of great masses of rock by a series of joints and cleavage planes, which had nothing whatever to do with the original stratification, but generally passed through it either perpendicularly or at a high angle—sometimes uninterruptedly for several hundred feet. The subject was very obscure, till elucidated by Professor Sedgwick; and its cause, especially as regards cleavage, is not yet clearly ascertained, though it is believed to be the result of some magnetic or electrical power, forcibly passing through the mass before its consolidation, and drawing or arranging the particles into one uniform direction, generally diagonal to the surface of the beds, and dipping at a high angle towards magnetic north. Cleavage is most distinct in the oldest sedimentary rocks, becoming gradually less so in the ascending series, till it entirely disappears in the younger or later formations; but it only affects the beds composed of fine particles, passing through coarse grits and conglomerates, without producing any change, and re-appearing in the alternating finer beds. Some beautiful examples of this occur in the Welsh rocks, connected with the paper he was about to read. The cleavage, where most perfect, generally destroys the original stratification, and renders it impossible to divide the mass in that direction. Common roof slate is a familiar illustration, which is always split in the direction of the cleavage. In some slates a streak or line of a different colour, often green, runs across it; this is the true line of bedding; but it is so completely obliterated, that no force, no art, and no length of exposure to weathering, would ever make the rock split in that direction. Joints have a very different origin from cleavage planes, and have originated in the shrinkage or contraction of the mass in passing from the soft into the solid state, exactly as the mud at the bottom of shallow pools splits and cracks in all directions after the evaporation of the water in summer. They are always most numerous and symmetrical in the older rocks, and most perfect in the finer grained ones, and are generally vertical to the surface of the beds. Their horizontal direction has definite, though unknown, connection with the same magnetic pole, for they are infinitely more numerous in that plane than in any other. They occur very copiously in coal, where they are more or less perpendicular to the surface of the beds, and may be recognised by their smooth shining sides, on which may often be traced fine parallel lines, which are sections of the true beds. This may be proved by the layer of soft charcoal which is frequently spread over them where exposed on the sides at right angles with the joints. The next set of joints and cleavage is full of interest, and ought to be more generally studied by practical observers; for it is impossible to arrive at any certainty in examining the older formations without a familiar acquaintance with it.

Mr. C. CLAY said, there was one class like those referred to by Dr. Black very strongly marked, near the town of Stalybridge, where the rock was disintegrated into very small portions, which assumed the form of curves; and the thinner these disintegrations, the more curve the appearances were. If it came to the shale, it was remarkably so; and, in the case of the road to Mottum, the shale was perforating three-fourths of the scale. He agreed with Dr. Black in thinking these appearances owing to the action of water after the body of rock had been raised from its original position.—Dr. BLACK said, one thing which led him to suppose that this disintegration was not owing to mere weathering, was the fact that the disintegrated bodies were in a very deep layer, perhaps two feet from the top to the bottom.—Mr. BOWMAN—Of course, the softer the rock the deeper the disintegration.—Dr. BLACK said, the rock was not so soft—it was as hard as roofing slate. The imitations were thin, from half an inch to an inch generally. The deposit above had been there for ages, being made, probably, at a period when the temperature of this country was high, compared with what it now is. These could have been so weathering since the tertiary period.—Mr. BOWMAN thought the frost could not have acted subsequent to the deposit.—Dr. BLACK said, the mere action of the water could never have shifted the pieces, and, therefore, the whole disintegration and removal must have taken place before that deposit was made. He mentioned that he had seen an example of this sort on the new Bolton and Preston Railway, upon the edge of the line.

**METALLURGICAL CHEMISTRY.**—A lecture on chemistry—chiefly metallurgical—was delivered by S. Mayle, Esq., of Burslem, at the Town Institution, on Monday week. The lecture commenced by speaking of the vast importance of the science of metallurgical chemistry to this country, and expressed his regret at the failure of the proposal for establishing a mining school, for instruction in this and the kindred sciences. He then reviewed the various copper ores, and salts of copper found in Cornwall, and described the principal tests for their detection. On these a great many experiments were performed; and there were also exhibited modes of extracting iron, lead, and silver, when held in solution. Having shown the means of detecting these metals, in chemical solutions, Mr. Mayle explained how their presence, as well as that of several other deleterious substances, may be proved in articles of food, strongly urging the importance of attention to this means of preserving health. The presence of several poisonous or injurious substances was shown by experiment, in water, and in adulterated bread, and wine.

## EXPLOSION OF STEAM-BOILERS ATTRIBUTED TO ELECTRICITY.

We mentioned, on a former occasion, in connection with the discovery of the development of electricity from effluent steam, that M. Tassin, an engineer in Belgium, had many years ago conceived the notion that electricity is the cause of the explosion of steam-boilers. We find, in a late Number of the *Fanal*, a more full account of the explosion of the steam-boiler which M. Tassin adduces as confirming his opinion. We translate the account as an instance of the extraordinary effects produced by the expansive power of steam; and as an instance also of ingenious, though, in our opinion, fallacious modes of accounting for these effects, by the assumed operation of other causes. Now that the discovery of the development of electricity by the escape of high-pressure steam is exciting much attention, the cause assigned by the ingenious writer in the *Fanal* may appear to be more feasible than it would otherwise have been considered. For our own part we attach no weight to it whatever beyond that of a curious speculation.

The notion entertained by M. Tassin, that gradual pressure cannot burst an iron vessel, will not be borne out by other experiments. It is very probable that a boiler may be so put together that the rivets and joints may be the weakest parts, but it is also possible for it to be so constructed that the iron will rend before the rivets give way. We have ourselves seen a vessel rent asunder by the gradual pressure of a force pump.

M. de Marotte possesses a large distillery at Vieux Waleffe, which is worked by a small steam-engine of 8-horse power, and, consequently, has a boiler in proportion. The boiler is a cylinder, four feet in diameter and eighteen feet in length, with a flat bottom, and through the middle of it there is a large fire flue. This form of boiler is much used in Belgium, and which gives general satisfaction.

Some minutes before the explosion, the proprietor, examining the manometer, which indicated two atmospheres and a half, desired the stoker to raise the pressure of the steam. He, however, replied that it was sufficient for the work that was required at the time. Everything in other respects was acting as usual, and there was an abundance of water in the boiler.

Scarcely had M. de Marotte reached his house before the whole distillery was destroyed by a tremendous explosion. The boiler was separated in two, on one side the internal flue, about fifteen inches in diameter; on the other side, the main part of the boiler to which the other half of the bottom was attached. The flue, and the half bottom of the boiler attached to it, threw down the base of a chimney nine feet thick, and the walls of four other different buildings that were in its course.

The main part of the boiler flew off in an opposite direction, passing, in the first place, through the wall of the engine-house, twenty-three centimetres thick—broke the iron beam of the engine as a cannon ball would have done—broke all the machinery that intercepted its progress—overturned the other supporting wall of twenty-three centimetres of bricks—and also the wall of a barn built of bricks, a metre in thickness. But what is still more extraordinary is, that this barn, of the dimensions of ten metres, and heaped up with sheaves of corn to the very roof, was also passed through by the main part of the boiler, which pierced the wall opposite, of equal thickness to the preceding one, but in which, however, it stuck. But this is not all. The bottom of the boiler, to which was attached a small portion of the flue, was blown from the body of the boiler, and killed three oxen in a stable which intercepted its progress; even this was not enough to stop the course of the fragment, which shattered two strong wooden posts twenty-three centimetres thick. The external wall of the stable did not arrest the progress of this part of the boiler, which dashed into the middle of the neighbouring pond.

All this was the result of electricity. With respect to the effect produced merely by the explosion of steam, a workman who was in a room immediately above the boiler was hurled through the roof to a very great height, and falling in the yard on a heap of sand was killed on the spot.

Another workman in the interior of the brewery, which was at a considerable distance from the boiler, and who was looking towards it, was thrown to a distance of fifteen feet against an open door by which he supported himself. This man declares that the boiler appeared surrounded by vivid lightning.

It is the province of natural philosophers to explain this phenomenon, which renders all the methods of security hitherto invented to prevent explosions unavailing, and particularly the one which has been adopted in Belgium, for the security of our neighbours, which consists in building a party wall two metres thick. We will hazard the following explanation:—Every decomposition or change of condition in a body evolving electricity, the evaporation of water must also produce it; but as steam-boilers are never entirely insulated, the electricity returns to the general reservoir, as soon as it is formed. But is it not possible that the great number of copper pipes in connection with the boiler in this distillery may have served as reservoirs of electricity that was accumulated within them? Is it not likewise possible that a workshop, paved with asphaltum, might have the effect of insulating all the tubes, boilers, stills, &c., which are used in a distillery, and that hence all these apparatus charged with electric fluid at the highest tension may have caused the terrible explosion, the consequences of which we have just related?

It must be observed, that as the discharge took place from the interior of the boiler, it was not the lightning alone that produced the effect, but the lightning armed, as it were, with the two portions of the boiler. The lightning alone certainly could not produce such effects.

If this hypothesis be sound, we must correct all our notions, and all the methods at present adopted for obtaining security against the explosion of boilers; for, according to what follows, it would appear that all such explosions are the effects of electricity, and are never occasioned by a progressive pressure, to guard against the effects of which such great and useless precaution is taken. Some experiments, made by M. Tassin, tend to prove this assertion. That engineer is now persuaded that it is impossible to burst a boiler by progressive pressure.

Having filled a globular boiler of sheet iron, a quarter of an English inch thick, with water, he proved it with a force pump to thirty-six atmospheres of pressure. Having arrived at that point the manometer began to fall, notwithstanding the pump was kept at work. He observed that a fine vapour exuded from every part of the boiler, and fell down like a mist. This effect is explained by the extension and opening of the different layers of iron, and the stretching of the holes of the rivets, which become oval during the expansion of the iron plates, and allow more water to escape than is forced in by the pump. M. Tassin perceiving that he could not burst his boiler made with thick iron plate, fixed over the man-hole of another boiler, by a number of screws, a plate of iron, only a line in thickness. When it was subjected to an enormous pressure, the iron plate became convex, the holes of the screws were enlarged, as in the preceding experiments, and the water escaped faster than it could be pumped in. He substituted a thin sheet of tinned iron, with the same result; so that this engineer is now convinced, that owing to the tenacity of iron, it is impossible for an explosion ever to take place from a gradual increase of pressure in a boiler. He is likewise of opinion that no explosion can result from the gradual introduction of water into an empty and red-hot boiler, because he has many times repeated that dangerous experiment, the effect of which has been only to produce a contraction of the surface when cooled. He, therefore, refers all explosions of steam-boilers, with very few exceptions, to electrical discharges. These experiments are said to have been made by M. Tassin two years ago.—*Inventors' Advocate*.

**JOINT-STOCK MANIA IN BELGIUM.**—It appears by Mr. E. Tonnens's new work, *Belgium*, that there has been a mania for joint-stock speculation in that country scarcely equalled by the bubbles of 1825 in Great Britain, and attended with equally ruinous results. It will scarcely be believed (though it appears to be a fact) that between 1833 and 1838, 150 or 160 companies actually invested 350,000,000 *l.*, or about 15,000,000*l.*, in speculations—for insurance, mines, machine making, public works, export associations, glass manufactories, sugar refineries, cotton and flax mills, printing, brewing, in short, every imaginable undertaking that could be described in scrip. The mania originated with some similar undertakings projected by the King of Holland, but which, being prudently conducted, were moderately successful.

**COAL IN INDIA.**—We learn, by the *Asiatic Review*, just received, that a large vessel (the *Cheraman*) is employed to convey coal from Mangal to Singapore, and that the coal is fully expected to answer; the colliery, it appears, is now in full operation.



## MINING CORRESPONDENCE.

## FOREIGN MINES.

## REAL DEL MONTE MINING COMPANY.

[The statement of accounts that appeared under this head, in our last, was that of the Bolanos Company; the error, however, would at once be perceptible to the reader, from the nature of the despatches that follow. We are happy to state, there has been a monthly increasing profit since October, and that the total losses of the company for the year do not exceed \$70,000.]

## BOLANOS MINING COMPANY.

[The following accounts were erroneously printed, in the Journal of last week, as those of the Real del Monte Company; we here repeat them, as the most satisfactory course for both companies.]

## General Summary of Accounts for 1840.

Date.	Quantity raised.	Value.	Costs.	Loss.
	Mos. or. ar.	rs. c.	rs. c.	rs. c.
January	4,548 4 2	41,550 3 0	45,396 1 6	4,845 8 6
February	6,536 4 6	57,442 6 2	67,195 3 2	14,752 8 0
March	4,647 4 6	39,344 8 2	51,463 6 6	12,119 8 0
April	4,696 4 2	33,812 7 6	50,407 6 6	16,594 9 0
May	4,984 4 2	38,286 1 2	54,932 6 3	16,646 5 0
June	4,180 7 7	35,308 1 0	51,577 0 6	16,269 7 6
July	2,749 2 4	22,607 3 1	40,978 1 1	18,370 4 0
August	3,589 1 2	44,490 3 1	63,681 0 3	19,190 2 2
September	4,496 6 2	37,094 3 6	55,394 3 6	18,300 0 0
October	4,889 3 6	48,585 0 7	67,613 1 6	19,028 0 7
November	4,629 6 4	38,195 7 5	54,248 2 7	16,053 3 2
December		[Not received.]		

## ENGLISH MINES.

## HOLMBUSH MINING COMPANY.

March 29.—I beg leave to inform you that Hithins's shaft is sunk to a depth of 51 fms. 1 ft.—ground not so favourable for sinking. In the 110 fathom level west the lode is six inches wide, of mundle and spar, intermixed with copper ore. The lode in the 100 fathom level west is still very productive, being 1 ft. 6 in. wide, and worth 35l. per fathom. In the ninety fathom level west the lode is 1 ft. 3 in. wide, and worth 12l. per fathom. The rise in the back of the eighty fathom level, against Hithins's shaft, is still in favourable ground. In this level, east of the engine-shaft, the lode is 1 ft. 8 in. wide, composed chiefly of mundle and spar. The lode in the eastern stopes, in the back of the eighty fathom level, is 1 ft. 4 in. wide, and worth 20l. per fathom. The lode in the western stopes, in back of ditto, is two feet wide, and worth 35l. per fathom. In the seventy fathom level, eastern stopes, the lode is twenty inches wide, and worth 30l. per fathom. The lode in the western stopes, in back of ditto, is eighteen inches wide, and worth 25l. per fathom. The cross-cut to Hithins's shaft, at the sixty fathom level, and rise in back of ditto, against Hithins's shaft, are still progressing favourably. The tribute pitches are still yielding good supplies of ore. In consequence of the samplers not attending on Friday, as appointed, the sampling is now deferred until to-morrow. F. PHILLIPS.

## WHEAL LERDS MINING COMPANY.

March 27.—Eighty Fathom Level East—Lode eighteen inches wide, composed of ore and spar. Sixty Fathom Level East—Lode eighteen inches wide, producing two tons of ore per fathom. Fifty Fathom Level East—Lode eight inches wide, producing one ton of ore per fathom. We have suspended the western levels for the present. We have set a cross-cut south, at the forty fathom level, and a wize to communicate from the seventy to the sixty fathom level. C. H. RICHARDS.

## TRESTOL MINING COMPANY.

March 29.—The lode in the thirty fathom level, west of engine-shaft, is about three feet wide, good tribute ground. The lode in the forty fathom level, east of engine-shaft, is one foot wide, tribute ground. The lode in the thirty fathom level, east of Williams's shaft, is one foot wide, tribute ground. The lode in the twenty fathom level, east of Williams's shaft, is small and unproductive at present. The part of the Mine Park lode we are driving on at the adit level, west of John's shaft, is one foot wide, producing spar, and a small quantity of ore. Tregillas's lode, at the same level, is one foot wide, unproductive. H. WILLIAMS. JOHN MORCOM.

## TRELIGH CONSOLIDATED MINING COMPANY.

March 27.—We have finished cutting the pit at the seventy fathom level, at Christor, and shall commence driving east and west on the lode immediately. In the sixty east the lode is large, producing a little ore. The sixty west is not yet clear of the disordered ground. In the fifty west the lode is four feet wide, of a favourable appearance, worth 7l. per fathom. In Good Fortune shaft we have a very promising lode in sinking; it is about eighteen inches wide, composed of spar, mundle, and ore. In the thirty four west the lode is two feet wide, with a good leader of ore in the end, worth 4l. per fathom. The twenty fathom level is improved; the lode is two and a half feet wide, worth 6l. per fathom. W. SINGOCK.

## WEST WHEAL JEWEL MINING ASSOCIATION.

March 29.—The ground in Buckingham's engine-shaft is still troublesome for sinking. In the fifty seven cross-cut south the ground is rather harder than last reported. The thirty west, on Tolemore lode, is worth 8l. per fathom. The thirty west, on the south lode, is worth 4l. per fathom. The twenty west, on this lode, has not been taken down during the week; the wize in the bottom of this level is worth 8l. per fathom. The deep adit west, on Wheal Jewel lode, is worth 7l. per fathom, and the rise in the back of the level continues worth 25l. per fathom. S. LEAN.

## ROSE-DOWN MINING COMPANY.

March 23.—We have not yet driven through the large lode cut in the deep adit some time ago; we are still passing through a large mass of chlorite, with abundance of mundle, and with a small portion of copper ore mixed in spar, &c.; it continues exceedingly hard, and, of course, our progress is very slow, which is quite unexpected; we have already driven on this lode twenty-one feet. As to the primary objects (the great copper lodes) we are decidedly of opinion they are lying before us. During the last two months six men have only progressed in the adit on account of the intersection of this unexpected lode, about seventeen feet; of course we advise by every means to persevere in extending the adit south. We sold yesterday about 115l. worth of tin, the particulars and amount will be duly forwarded you. The appearance of the tin lode is such as we consider will pretty much more than pay its way for working—a statement also of which relative to the last return will be forwarded to you. R. ROWE.

## UNITED MILLER MINING COMPANY.

March 31.—Adit East—Lode about 9 ft. 6 in. wide, producing a little ore, with a promising appearance. Adit East West—Lode eighteen inches wide, poor. Ten Fathom Level—Lode three feet wide, producing a small quantity of ore. Twenty Fathom Level—Lode two feet wide, with but little ore. Thirty Fathom Level—Lode three feet wide, two feet of which is ore. Thirty-six Fathom Level—Lode three feet wide, producing some ore, but coarse in quality. Forty Fathom Level—Lode three and a half feet wide, ore throughout, but of coarse quality. Fifty Fathom Level—In the eastern end of this level the lode is three feet wide, with some stones of ore; western end three and a half feet wide, ore throughout. Sixty Fathom Level—In the eastern and western ends of this level the lode is four feet wide, one foot on the north part good ore. C. PENROSE.

## REDMOOR CONSOLIDATED MINING COMPANY.

March 24.—The engine-shaft is sunk to 10 ft. 9 in. below the fifty fathom level; the ground there is very easy for sinking—we have given 5l. 10s. per fathom; we consider, however, it will require the suspension (also in wize) from a week to ten days to make such alterations in the pit work as we require, to sink for the intended sixty fathom level. We have also set the fifty fathom level cross-cut to drive south by six men as a new bargain; it is taken at 3l. per fathom; the ground is very favourable. At the forty fathom level we have cut the south lode, it is about 1 ft. 6 in. wide, composed of copper ore, mixed in mundle, spar, and capel, but not rich for the former; the ore is of good quality, and has a promising appearance, we consider rather better than where it was cut at the thirty fathom level over, but as very trifling has yet been done, merely passing through it in the cross cut, that it is quite premature to attempt to form a correct opinion how the lode at this level will turn out. At the thirty fathom level we are driving west on a lode, which we have termed the middle copper lode; we have driven about six feet on its course, and find it to be eighteen inches wide, with an abundance of mundle; we consider this lode is deserving, and ought to be prosecuted. At this level we are also driving south on the outer lode, the prior we give for driving is 35s. per fathom; the lode is in short furnace nature, about six inches wide, and the back of this level will set at about one-half cut of 1l. During the past month we have raised a wize through from this level (thirty), to the twenty fathom level, in doing which it has proved of great advantage in ventilating the said levels, as well as discovered some tribute ground on the copper lode, which we have been able to set a new pitch at 7s. 6d. out of 1l. At Hurl Down adit, going south, we have driven nearly six fathoms—ground is moderate, price given to-day is 5s. per fathom. We have about eight tons of silver-lead ore broken, and the computed quantity of copper is twelve tons. R. ROWE.

## YANAN SILVER-LEAD MINING COMPANY.

March 28.—Twelve pitches have been set, and also others, their taking not expiring—making the number of pitches working altogether 31, varying from 2s. to 14s. out of 1l., on the value of the lead only, as usual. During the past month we have been thrown back in raising the regular quantity of ore, on account of not being able to work two of our principal pitches in the back of the 105 fathom level, the air becoming so impure obliged us to suspend the working of the said pitches for a fortnight, and, in the interim, was under

the necessity of employing the two parties of tributaries to communicate a wize from the 95 to the 105 fathom level, for ventilating that part of the mine, consequently our next sampling will not be quite so much as we are accustomed to sample. This work, however, I am glad to say, is now completed, and the tributaries alluded to are gone to work regular, and have fair prospects of breaking a pretty large quantity of work. The several other pitches remain stationary as to prospects since my last inspection of the 25th ult. At the 135 fathom level going south of the engine-shaft the lode is 1 ft. 6 in. wide, mixed with ore, spar, and capel. At the 125 fathom level the lode is one foot big, yielding a little ore. At the 115 fathom level the lode is six inches wide, and rich for ore. In the 105 fathom level the lode is heaved to the east of its regular course, on account of a furnace intervening, and we expect to drive about 12 ft. westward to cut the main lode. In the 95 the lode is three feet wide, very hard, producing good saving work. We have suspended for the present the eighty-five and seventy-five fathom levels, and have employed the same men to sink a wize from the latter to the former level, for the purpose of dividing the ground for tribute, and as well improving the air in that part of the mine. We expect in a week to hole the wize rising from the forty-five to the thirty-five fathom level, and, when effected, will enable us to carry on our tribute there with greater facility. The last computed four-four tons of ore was sold on the 15th inst. to Messrs. R. and W. Michell, at 16l. 11s. per ton; and the computed fourteen and a-half tons to B. Somers, Esq., at 15l. 11s. per ton. R. ROWE.

## PRODUCE OF THE PRUSSIAN MINES DURING THE YEAR 1837.

[From the <i>Bulletin de la Société d'Encouragement</i> .]					
Coal	..... quintals.*	41,370,961	Sulphur	..... quintals.*	486
Borax coal (lignite)	.....	16,450,000	Silver	.....	8,824
Oxide of manganese	.....	5,852	Lead	.....	23,902
Galenite (sulphuret of lead)	.....	42,739	Oxide of lead	.....	18,847
Cast-iron	.....	1,945,882	Copper	.....	19,847
Bar-iron	.....	1,141,535	Wrought copper	.....	18,145
Sheet-iron	.....	1,033,573	Brass	.....	18,344
Steel	.....	103,338	Zinc	.....	216,486
Refined steel	.....	42,463	Roller zinc	.....	18,838
Alum.	.....	8,791	Alum	.....	48,577
Armenian acid	.....	3,178	Sulphate of iron	.....	33,807
Sulphate of antimony	.....	328	Sulphate of alumina	.....	3,457
Regulus of antimony	.....	275	Common salt	.....	41,109

[The above account is valuable, as giving an idea of the mineral produce of Prussia, though we have reason to believe that the statement, in some respects, is not exactly to be depended upon, being under the actual quantity yielded by the mines.—Ed. M. J.]

\* The quintal may be assumed at about 123 lbs. English.

## GREAT COAL-FIELD, NEW BRUNSWICK.

The Great Coal-field is situated between the primary rocks of the county of Charlotte and King's County, and the Straits of Northumberland, on the Gulf of St. Lawrence. Only the south and south-east sides have yet been explored; the west, north, and north-east sides still remain to be examined, and its limits, therefore, in the latter directions, yet remain unknown. The division of this coal field, situated southward of the St. John, is the segment of a large circle, described between the Keswick above Fredericton, and the Oronoto below Gasquetown, and touching at Skin Creek and the head of the Oronoto. Its south-eastern side extends along the trap and syenite rocks of Springfield, and the dividing line between King's and Queen's, Westmorland and Kent counties, to the Straits of Northumberland. From one of the branches of the Oronoto to the Saint John, and from thence eight miles eastward of the entrance of the Washademoak, the old red sandstone and carboniferous limestone appear, cropping out from beneath the millstone grit, along a distance of upwards of thirty miles. These formations have been already described. From what I have been able to discover, I believe that this coal-field extends in a northerly direction to Bathurst, a distance of 150 miles, and to Miramichi, 120 miles, and from the latter place along the coast to Shediac, which may be estimated at seventy miles. Until the north-east side of this vast coal tract is explored, it would be impossible to give a correct account of its area; but it may for the present be considered equal to 5000 square miles!!! We are aware that in making this statement, we must necessarily be exposed to remark; but it is nevertheless supported by the most unquestionable facts; and we have only to appeal to them, in vindication of what is here recorded. This tract may, perhaps, bear the reputation of being the largest coal-field ever discovered on the globe. Over the whole of this vast area, the conglomerates, sandstones, shales, ironstone, and frequently coal, appear at the surface, filled with innumerable remains of plants, that have long since ceased to exist; but whose relics, as they are seen in almost every rock, bear ample testimony of the herbage of former periods. This vast expanded tract, in every part, abounds in tropical plants, many of which have evidently been changed into enduring beds of coal, while others have been converted into different kinds of mineral matter, and form the most faithful record of the changes this earth has undergone since it first came from the hands of its Supreme Architect. To distinguish this extensive tract from the Westmorland district and other coal-fields in the British provinces, we have designated it by the name of the "Great New Brunswick Coal-field," which, for its magnitude and wealth, will be better known long after its first geological pioneer has ceased to travel over its surface.—Geological Survey of New Brunswick.

## MINING NOTICES.

[Under this head we purpose collecting such paragraphs as may appear in the provincial and other Journals, having reference to discoveries and improvements in mining operations at home and abroad. It is hardly necessary to observe, that we must not be considered to admit the correctness of the information conveyed, which, in too many instances, requires cautious investigation—the sanguine expectations of parties in some instances, and the want of honesty in others, throwing a degree of responsibility on a Journal in giving publicity to reports, which we do not intend taking upon ourselves.]

COAL MINE, QUEEN'S COUNTY.—The coal mine on the lands of Deer-park, near Monmouth, which was in progress of being opened last summer, but relinquished through the winter, has been recommenced on Monday, the 15th inst. The proprietor of the soil (Sir C. H. Coote, Bart.) has procured several skilful working miners, and the necessary implements for working, and it is generally expected that ere long he will be repaid by a most abundant bed of coal.—Leinster Reformer.

DIAMOND MINE IN SUMATRA.—The last letters from Sumatra state, that a mine had been discovered in the district of Dolandou, which appears to be as rich as those of Borneo. The gold mines of Borneo and Kampong Kardi, which have been worked since 1837, appear to be more productive in proportion to the depth from the surface.—Athenaeum.

IRON IN NORTH AMERICA.—We learn that a considerable change in our extensive dealings with the United States, in the article of iron, is likely to take place, and the enormous exports of pig-iron to that continent will be greatly diminished, in consequence of late discoveries of immense quantities of iron of a very superior quality in Maryland. What renders this discovery of so great value is, that both water and railway carriage are at hand, and an inexhaustible supply of coal and wood. Mr. J. E. Piddick, the agent for the estates, has dispatched an experienced mining engineer and a surveyor to report on this property, which we hear is to be worked on a large scale by English capital.—Railway Magazine.

IMPORTANT DISCOVERY.—Mr. Neville, of the Customs, has discovered a metal, supposed to be platinum, in the bones of a rock about eighteen miles distance from Melbourne. It is heavier than iron, being eleven times more weighty than water. Several chemists have been attempting to analyse it, but without effect. Mr. Neville says that he could load several ships with it, in such quantities has he seen it. It is somewhat singular that Mr. Neville discovered the same metal, but not of so pure a quality, on the Sydney side of the country, some months ago, and as the chemists could not analyse it there, he sent it home to England, but has not yet heard of the result of his speculation.—Port Phillip Herald.

DISCOVERY OF COAL IN FRANCE.—An important discovery of a mine of pit coal, which presents a seam of five metres thick, has just been made in the neighbourhood of Quimper, a short distance from the port of Brest.—Moniteur Industriel.

COAL MINES.—Coal has been obtained near Patricroft, in Lancashire, on the land of T. J. Traff, Esq., at the depth of 1350 feet from the surface. "We have not heard of a shaft before (observes the Manchester Times) equal in depth to this." We inform our contemporary, that Messrs. Pemberton, in 1834, reached a seam of coal in Monkwearmouth at the depth of 1575 feet below the surface; and have since sunk to a still greater depth, their mine being not, we believe, the deepest in the world below the level of the sea, but the deepest below the surface of the earth.—Gateshead Observer.

## MINE ACCIDENTS.

St. Hilda's Pit, South Shields.—On Wednesday week, Luke Laidler fell down St. Hilda's pit shaft, and was instantly killed.

Team Colliery, near Barnsley.—A fatal accident occurred at this colliery by the falling of a large stone from the roof of the mine, it struck two men, killing one, and breaking the leg of the other, besides inflicting other serious injuries.

Concise Mine.—On Friday week, John Gill met with his death by falling from the 105 to the 220 fathom level, in Concise Mine.

Upper Band Works, Barnsley.—Two copper-ore were seriously smothered at these works last week; they were engaged in tapping the furnace, when an explosion took place.

Explosion Colliery.—As G. Goodell was going to his employ at this colliery, he fell into a deep pit, and was killed on the spot.

## ANTHRACITE COAL TRADE OF THE UNITED STATES.

The following table exhibits the quantity of anthracite coal sent to market from the different regions from the commencement of the trade in 1820 to 1841, together with the annual increase and consumption:—

Years.	Schuyll kill.	Mauch Chunk.	Braver Mead.	Hants. ton.	Sugar Loaf.	Pine Grove.	Shamokin.	Lackawanna.
1820	—	363	—	—	—	—	—	—
1821	—	1,070	—	—	—	—	—	—
1822	—	2,240	—	—	—	—	—	—
1823	—	3,803	—	—	—	—	—	—
1824	—	5,341	—	—	—	—	—	—
1825	3,309	26,308	—	—	—	—	—	—
1826	16,816	31,180	—	—	—	—	—	—
1827	29,493	32,074	—	—	—	—	—	—
1828	47,181	30,532	—	—	—	—	—	—
1829	78,293	28,110	—	—	—	—	—	7,009
1830	89,384	41,750	—	—	—	—	—	45,000
1831	81,804	49,386	—	—	—	—	—	84,000
1832	209,371	75,000	—	—	—	—	—	111,777
1833	230,388	125,000	—	—	—	—	—	117,777
1834	228,692	106,244	—	—	—	—	—	43,700
1835	333,685	131,280	—	—	—	—	—	90,000
1836	443,734	146,302	—	—	—	—	—	100,000
1837	335,289	102,380	20,617	—	—	—	—	115,397
1838	434,681	152,099	44,976	16,221	—	—	—	78,331
1839	442,698	145,631	58,479	34,000	7,380	30,510	11,810	127,360
1840	432,791	102,164	43,619	50,300	28,000	38,000	15,000	146,470
	3,779,769	1,173,478	100,931	190,387	36,380	78,400	37,433	1,002,125

Years.	Aggregate.	Annual increase.	Consump.
1810	305	—	—
1821	1,070	—	—
1822	2,240	—	—
1823	3,803	—	—
1824	5,341	—	—
1825	33,009	—	—
1826	35,000	—	—
1827	41,362	—	—
1828	77,473	—	—
1829	116,460	—	—
1830	174,794	—	—
1831	176,800	—	—
1832	209,371	—	177,000
1833	230,388	—	218,000
1834	228,692	—	43,000
1835	333,685	—	84,000
1836	443,734	—	100,000
1837	335,289	—	115,397
1838	434,681	—	78,331
1839	442,698	—	127,360
1840	432,791	—	146,470

The above table includes the shipments from all the anthracite regions in the State, except the Wilkesbarre Basin, from which we have no returns. This supply, however, does not affect the Atlantic market—as the whole quantity, which, in 1837, was 17,492 tons, is consumed in the interior. The new impetus given to the coal trade by the erection of anthracite iron works on the Susquehanna, will greatly increase the supplies from the Wilkesbarre region hereafter.

## IMPROVEMENTS IN STEAM-BOILER FURNACES.

The following notice of some improvements lately adopted in the construction of the furnaces of steam-engine boilers, by M. Denicot, son, an engineer and machinist, is given in the French publication *L'Esper*. A more particular description of the nature of the improvement, by which the effects stated are produced, is promised in a succeeding Number.—Until the present time, almost all the furnaces of steam-engines have been constructed on the same plan; the same false and injurious principle is universally adopted; they are constructed on one system, and, it might be said, they are all the work of one boiler, so little difference is perceptible between them. Among the number of defects remarkable in the present furnaces, resulting from their defective construction, we will particularise the following. The furnaces communicate too much heat to a certain portion of the heating surface of the boilers, and communicate scarcely any heat in the rest of the surface, though the heat ought to be equally distributed—hence the frequent repairs requisite in the parts of the boilers that have been too much heated, and the increased danger of explosions. The caloric effect of a portion of the combustible gas is destroyed by its not being entirely consumed; there escapes up the chimney also a considerable quantity of coal, in a state of almost impalpable powder, carried away by the draft without being burnt—hence arises that dense black smoke which annoys from a distance the presence of a steam-engine, and which is, also, too often the cause of complaint and annoyance to those in the vicinity. They consume too much coal, and do not make the most of that which is used. On the present plan of constructing furnaces it is not possible instantaneously to intercept the communication of the flame with the boiler, which on many occasions would be very advantageous.

We can instance, in support of our assertion, the steam generators of the chief manufactory in the department of Haut-Marne. They have scarcely worked from 500 to 600 days of twelve hours each, and have had part of their surfaces renewed nine times. The furnaces which heat these generators must consequently be very bad, when their operations are attended with such results. These engines also consume five kilograms of coal an hour for every horse power, which is double the quantity that a good engine ought to consume. We will notice, in passing, another steam-engine at the same establishment, placed at the mouth of a blast-furnace, and heated by the flame that escapes. In this case, notwithstanding the quantity of heat which may escape from the gases at the furnace mouth when burning, these gases have never been made to communicate heat to the boiler. They become, in short, entirely extinguished when brought into contact with the pipes of the boilers, and are not relighted till they reach the chimney, the walls of which they make red-hot.

So little has the management of this kind of furnaces been understood, that the engine has been taken to pieces, and altogether stopped, rather than attempt those necessary improvements which common sense suggests should have been made.

The arrangements which we have made for generators, heated either by flame direct from the fuel, or by the flame of gases escaping from blast-furnaces, have enabled us to attain the following results:—The whole heating surface of the boiler is raised to the same temperature, whatever the length of the boiler may be. The generators last a considerably longer time, in consequence of not being burnt in places as they formerly were. Being no longer exposed to the flame of a furnace of a high degree of temperature, they are not in danger of becoming red-hot, and the chances of explosion, if not impossible, are extremely rare. All the combustible gases are completely relighted and burnt; they not become extinguished till their heat is exhausted, and they are converted into combustible gas. Coal dust is also entirely consumed by the burning gas, and no dense smoke is seen issuing from the chimneys, which it is no longer necessary to build to so great a height.

All the gas and coal being turned to account, the saving in the consumption of fuel is considerable. The heat underneath the generators may be diminished without extinguishing the fire, which is of advantage in cases when, owing to a mistake, the water in the boiler is too low. It is then possible to let it cool, and avoid the explosion which usually happens from this cause.

The plan on which our furnaces are constructed is remarkable for its simplicity, any experienced builder being able to execute them.

## IMPROVED DIAL OR COMPASS NEEDLE FOR MINES.

At the quarterly meeting of the Manchester Geological Society, held at their rooms, on Thursday week, Mr. Hutton announced that he had received from an old mineral hunter and working miner (named James Hamden, living in King-street, Oldham), a letter on the subject of the compass needle, stating that, having been a long time engaged in making mineral surveys, he had experienced the difficulties arising from numerous interfering metallic bodies, which tended to overpower or lead the needle from its natural direction, so that it could not always be depended upon, and, consequently, subterranean surveys would not always coincide. The rails, and particularly those of beaten or rolled iron, attracted and repelled the needle; water, holding a quantity of iron in solution, and various metallic bodies, interfered with it. To obviate these evils, the writer had invented a compass needle (one of which accompanied the letter, as a donation to the society), which he described as consisting of three parts—viz., two needles or bars of steel, each 14 inch long, and each of seven bars having a north and a south pole, divided in the middle by a brass bar, which separates the north pole of one needle from the south pole of the other by about 14 inch; and the writer states, that the efficacy of one of these for the other is so strong, that the priority of the needle, as a whole, is maintained—that neither water, ironstone, nor any substance would set upon it—that the rails could not affect it, nor even the friction of the iron wheels of waggons on the rails, though this would draw the compass needle five points from its "natural" position. He stated north or south pole of his needle could not be led away five or six right directions, because of the union and mutual attraction of the two lateral poles. Were this property of his needle considered, he thought it would improve itself into general use.



# STANNARIES OF CORNWALL. IN THE VICE-WARDEN'S COURT. BORLASE AND OTHERS v. THOMAS.

**WHEREAS** the Vice-Warden d.d. on the 27th day of January last, order that a sale be made of (amongst other things) the machinery and materials upon, and belonging to, Wheal Rose Mine, in the parish of Sithney, within the said stannaries, under the direction of the registrar of the court, and that the proceeds of such sale should be applied by the said registrar in the manner directed by the decree, in the above mentioned cause;—Notice is hereby given, that, pursuant to the said decree, a PUBLIC AUCTION will be held at WHEAL ROSE MINE aforesaid, on Tuesday, 25th of April next, at Eleven o'clock in the forenoon, for selling either together, or in lots, the undermentioned MINING MACHINERY, MATERIALS, and other effects—viz., a capstan and shears, two horse whins, with shaft tackle, twenty fathoms of ladder, a quantity of detenture, whie, half, and quarter balk planks, &c., a timber shed, three pieces of Memel red timber, three large iron blocks, a large quantity of wrought and cast-iron, smith's bellows, anvil, vice, grindstone, carpenter's bench, wine, kibbles, hilt, chisel, &c., &c., a quantity of brick, nails, about twenty dozen of candle, consisting of home furniture, &c., &c.

For viewing the same, application may be made at the mine, and for further particulars (if by letter post p. id.) to Mr. T. P. Tyacke, solicitor, Helston; or to Messrs Paul and Roberts, solicitors, Truro.

Dated the 21st day of March.

**WANTED**, in a locomotive and other engine manufactory, in one of the principal towns in the kingdom, two respectable youths as AP- PRENTICES. As the utmost care and attention will be bestowed in instructing them in every branch necessary to skill and competent knowledge of the business, a commensurate premium will be expected. Letters addressed "Mechanics," and sent to the office of this Journal, will be attended to after the 1st May.

## RAINBIDGE ON MINES AND MINERALS.

Just published, in demy 8vo., price 15s. boards.  
**A PRACTICAL TREATISE ON THE LAW OF MINES AND MINERALS**, comprising a detailed account of the respective Rights, Interests, Duties, Liabilities, and Remedies of Landowners, Adventurers, Agents, and Workmen; and of the Local Customs of Derbyshire, Cornwall, and Devon, with an Appendix of Legal Forms, relating to Grants, Leases, Transfers, Partnerships, and Criminal Proceedings, and a General Index.

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" A work of much importance and interest, and highly creditable to the talents and industry of the author, as furnishing information which could not be acquired by an intimate connection with the mining districts, as well as a legal knowledge of the several points discussed. A work which must be popular, and ought to be in the hands of every miner, and to be in the office of every solicitor, while its usefulness to the capitalist and mine adventurer, will be acknowledged even by a brief review of its contents."—*Mining Journal*.

Henry Butterworth, Fleet street, London; M. A. Richardson, Pilgrim street, Newcastle-on-Tyne; and all other booksellers.

## THE CIVIL ENGINEER AND ARCHITECT'S JOURNAL.

SCIENTIFIC AND RAILWAY GAZETTE, price One Shilling and Sixpence.—No. 45, for April, contains Drawings and Description of Towing Paths and Banks of Canals in Great Britain—Railways Bill—Engineering Works of the Ancients, No. 2—Episodes of an Establishment for procuring Models from Monuments of Art—Harbours on the South Eastern Coast—Coles's Patent Steam Axis Tree—Railway Statistics—Remarks on the Central Forces of Bodies revolving about Fixed Axes—On the power of Fluids in Motion—Roman Architecture—Architecture of Liverpool—Railway Accidents—Public Safety and Convenience—Candidates' Note Book, Fasciculus 25—Chertem Church, &c.—On the Curvature of the Arches of the Bridge of the Holy Trinity—New Inventions and Improvements—The Artesian Boring at Paris—The Magin Lighthouse—The Plymouth Breakwater Lighthouse—Mersey and Irwell Navigation (Mr. Palmer and Mr. Bateman's Reports)—Warming Buildings by Hot Water—On the Style of Wren—Reviews of New Books—Proceedings of the Institution of Civil Engineers, Royal Institute of British Architects, and the City of Architectural Society—List of New Patents—and numerous professional articles and remarks.

London, Pall Mall; R. Groombridge, Panzer alley, Paternoster row.

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## MEETINGS OF SCIENTIFIC BODIES.

### IN THE ENSUING WEEK.

SOCIETY.	PLACE OF MEETING.	DAY.	HOUR.
Royal Asiatic	14, Grafton street	Saturday	2 P.M.
Westminster Medical	Exeter Hall	Saturday	8 P.M.
Royal Geographical	21, Regent street	Monday	8 P.M.
Linnean	Robinson square	Tuesday	8 P.M.
Horticultural	21, Regent street	Tuesday	2 P.M.
Civil Engineers	25, Great George street	Tuesday	8 P.M.
Architectural	25, Lincoln's Inn fields	Wednesday	8 P.M.
Society of Arts	Adelphi	Wednesday	7 1/2 P.M.
Geological	8, Museum House	Wednesday	8 P.M.
SI. Society of Literature	St. Martin's place	Thursday	4 P.M.

## PUBLIC COMPANIES.

### MEETINGS.

Bahia Steam Navigation Company	George & Vulture Tavern	April 3	11 1/2
Ashby de-la-Zouch Canal Company	Ashby de-la-Zouch	3	11
Argus Life Assurance Company	Throgmorton street	6	2
Blaenavon Iron and Coal Company	London Tavern	6	1-2
River Don Company	Manchion house street	7	11
Britannia Life Assurance Company	Pincon street, Bank	8	12-1
Imperial Gas Light and Coke Co.	25, John street, Bedford row	8	1
Chesnut and Blackhead Railway	Chesnut Rooms, Liverpool	10	12
Maryport and Carlisle Railway	Adelphi	10	11
North Wales Railway	George & Vulture Tavern	14	1
Newcastle on Tyne & Carlisle Rwy Co.	66, Close, Newcastle	22	12

## CALLS.

East Tretford Mining Company	24, April 16	Barclay, Brown, and Co.
Wheal Looe Mining Company	24, April 18	Barclay, Brown, and Co.
Pulbrevon Mining Company	24, April 14	Bosmanquet and Co.
South Australian Company	24, April 13	Ladbroke and Co.
London and South Western Railway (Gospel Branch)	24, April 18	Williams and Co.
Bombay Dock Company	24, April 18	Williams and Co.
Bedfordshire Mining Company	10, April 21	Bosmanquet and Co.
Great N. E. of England Railway	10, April 21	J. Pease, Darlington.
Agricultural and Commercial Bank of Ireland	10, April 21	Office.
Hartford Dock and Railway	24, April 21	Barnett and Co.
Wheal Wattle Mine	24, April 19	Manchester & L'pool Dist. Bk.
Hungford and Lambeth Dock	24, April 18	London and County Bank.
London and County Bank	24, April 18	Shoe, Martin, and Co.
London and County Bank	24, April 18	London Joint Stock Bank.

## DIVIDENDS.

Chesnut Copper Mines	12 per share	25, April 5
United Mine Company	10s per share	Office, Adam's court
Commercial Bk. of New Orleans	4 per cent.	Bank, Irving, and Co.

## JOINT STOCK BANKS.

Birmingham Banking Company	3 per cent.	per share
Bank of India	4 per cent.	per share
Bank of London	4 per cent.	per share
Bank of Manchester	4 per cent.	per share
Bank of Commercial	4 per cent.	per share
Bank of Liverpool	4 per cent.	per share
Bank of North Wales	4 per cent.	per share
Bank of Scotland	4 per cent.	per share
Bank of South Wales	4 per cent.	per share
Bank of the West	4 per cent.	per share
Bank of the East	4 per cent.	per share
Bank of the North	4 per cent.	per share
Bank of the South	4 per cent.	per share
Bank of the West	4 per cent.	per share
Bank of the East	4 per cent.	per share
Bank of the North	4 per cent.	per share
Bank of the South	4 per cent.	per share

## RAILWAYS.

London and Greenwich	7s per share
Liverpool and Manchester	4 per share
North Wales	4 per share
South Wales	4 per share
West of England	4 per share
York and North Yorkshire	4 per share
York and North Yorkshire	4 per share
York and North Yorkshire	4 per share

## GOLD AND SILVER.

Foreign gold in Bank	100 shillings	per oz.	48 1/2
Foreign gold in Bank	100 shillings	per oz.	48 1/2
Foreign gold in Bank	100 shillings	per oz.	48 1/2
Foreign gold in Bank	100 shillings	per oz.	48 1/2
Foreign gold in Bank	100 shillings	per oz.	48 1/2
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Foreign gold in Bank	100 shillings	per oz.	48 1/2
Foreign gold in Bank	100 shillings	per oz.	48 1/2

## NOTICES TO CORRESPONDENTS.

The importance attached to the properties of the sulphur ores of this country, and their application to purposes which, until of late, were unknown, renders a paper, inserted in another column (with which we have been favoured by a correspondent), of peculiar interest at the present moment, it affording a comprehensive view of the nature and properties of sulphur, and, among other subjects treated upon, one which will excite particular attention—viz., the extraction of sulphur from pyrites, with remarks on the useful employment of the residue of distillation. We shall be most happy to receive further communications on this subject from our correspondents.

**GEOLOGY OF THE PARIS BASIN.**—We have in type an interesting Geological Description of the Paris Basin, but are compelled to postpone its insertion.

**BARRIS IRON COMPANY.**—We have too many claims on our columns to admit of devoting so much space as our Liverpool correspondent requires, to "place the company's present position in the true light before the public, and how it might have been averted." We must publish a pamphlet, if he wishes to give publicity to his views, unless he can express them in a paper of a moderate length.

The papers on the Resources and Properties of Coal, and the Geological Structure of the Wealden District, are unavoidably postponed.

**INSTRUCTIONS FOR BLASTING.**—The concluding part of this paper will be published next week.

We have been favoured by a correspondent ("A.") with a sketch and description of an improved Machine for Trunking "Slime," which shall appear in our next.

**DURHAM COUNTY COAL COMPANY.**—The publication of "T. M.'s" letter had better be deferred until after the forthcoming meeting. Should the desired information not then be afforded, we shall publish it, together with several others, which we have withheld, in the hope that the directors will see the necessity of meeting the shareholders with candour, and render a just account of all their transactions.

It will afford us pleasure to forward the case for the opinion of "K. A. T. E."

# THE MINING JOURNAL, Railway and Commercial Gazette.

LONDON, APRIL 3, 1841.

It is with pleasure we refer to our columns of the present Number, as well as to those which have anticipated them since the commencement of the year, as evidence of the increasing interest excited by the matter which finds place of record in the MINING JOURNAL. We are not naturally egotistical, nor ought we to be so considered on the present occasion, as it is to our correspondents that we are indebted for being able to submit to our readers a weekly periodical possessing so many claims to the attention of the geologist, the miner, the collier, the assayer, the engineer, the scientific reader, and the adventurer, while the "postscript" affords to the fundholder and the commercial man a review of the week's proceedings in the money market, and the latest intelligence.

We are induced to direct attention to the subject-matter of the Journal, in the present instance, for the purpose of availing ourselves of acknowledging our obligations to several correspondents, whose papers have lately appeared in our columns, and which we have illustrated by diagrams, being anxious at all times to render the MINING JOURNAL useful and instructive, and, indeed, to carry out the objects which its very title indicates—a Journal of Mining Operations—leaving it to our readers to ledgerise the several articles by double or single entry, as they may think fit. In referring to the able articles on the working of collieries, and operations connected therewith, the papers of Mr. M. DUNN, of Newcastle, Mr. STANLEY, of Sunderland, and other correspondents, who have subscribed themselves under initials, or fictitious signatures, we have more particularly to acknowledge, from the interest which has been manifested on this subject. The communications of Mr. BYRNS, Mr. M. J. ROBERTS, Mr. PRIDEAUX, and others, on the various modes of assaying—the letters of Mr. BUDGE and others, on dialling and underground surveying, with the numerous subjects of a scientific character treated upon in our columns, derived from correspondents, have materially enhanced the character of the Journal. The paper in our present Number, "On the Application and Properties of Sulphur," amongst other original papers which have from time to time appeared, will be read with interest, and at once assure our subscribers that we are not less anxious to render information, from the best sources, than they are to acquire it.

The general meeting of the Cambrian Iron and Spelter Company, held on Wednesday last, affords opportunity for making some few remarks on the course adopted in excluding the press—the object of which is too apparent to render observation necessary, except in cautioning shareholders in other companies from being made the dupes of a secret conclave. We have, on more than one occasion, animalvered on the system of secrecy observed, and have not hesitated to express our conviction, that there were good and sufficient reasons—at least, in the opinions of the directors—why the public should not be put in possession of the affairs or circumstances of the company, or the absent shareholder made acquainted with the nature of the discussion which might take place at the general meetings. The mere printing of a report and accounts is, in a great measure, a farce, even in the cases where such course is adopted, compared with the interrogatories and explanations afforded at a public meeting. This company, however, may lay claim to the right of exclusion of the press, perhaps, more than others we could name, for, if we are rightly informed, the scheme may be considered to have been, in a great measure, a failure—the public would not "bite," and hence the necessity of its being upheld, in hopes of better times, by a few individuals. Backed by the support of the coffers of the London Joint-Stock Bank, much has, doubtless, been done in the expenditure of money, but we should question whether the profits of the company, when in effective operation, will even pay the interest of the money they may have, or will find necessary, to borrow. We should think nearly 200,000*l.* has been already expended on the concern, which is highly creditable to Mr. BUNYON, the engineer, for the erections and general laying out, without regard to the cost incurred. We will not, however, dwell on this particular company, but proceed to enumerate one or two other Iron Companies, the directors of which evince equal jealousy of publicity being given to their proceedings. We will first begin with the British Iron Company. This company, projected in the year 1824 or 1825, has, since that period, called up its subscribed capital no less a sum than 1,200,000*l.* and by the quoted prices, the shares are worth (7) 50,000*l.* less than nothing—some 300,000*l.* or 400,000*l.* have to be paid Mr. ATTWOOD—legal proceedings are threatened by the directors against the shareholders in default, and, on the other hand, an application to the Court

of Chancery by the latter against the board of directors—this is example the first of the "silent system." The next upon our list is the Rhymney Iron Company, the capital subscribed on the shares in which is 500,000*l.*, and which is represented in the market at 50 per cent. discount. Next, the far-famed Talacre Iron Company, of which we have already said enough for the present, although much remains behind. Here, then, we have enumerated four Iron Companies, which preserve silence, and do their deeds in darkness; the only exception in our share list being the Blaenavon Iron Company, to the meetings of which we have at all times had ready access, and whose shares maintain a fair price in the market, attributable alone to the candour and openness manifested by them, which engenders public confidence, and ensures public support. It will be seen that the companies we have enumerated stand thus:—British Iron Company, amount paid, 1,200,000*l.*—value 80,000*l.* less than nothing, or a premium given by the holders of the shares of 80,000*l.* over and beyond the money subscribed, with the view of avoiding further calls and responsibility. Rhymney Iron Company, amount paid, 500,000*l.*—value 250,000*l.*, and unsaleable to any extent. Cambrian Iron Company, amount paid something like 200,000*l.*, with no market value; and the Talacre Iron Company, the shares in which, even if paid up in full, no one would take as a gift. Such is the state of these companies, which refuse to give publicity to their proceedings, the cause of which may be well guessed at, while the effect is seriously felt by the proprietary—yet who have only themselves to blame. If among other companies we wished to adduce another instance, we might take the General Mining Association, brought out under most splendid auspices, and the shares of which obtained high premiums; this company, after having put forward reports, not only calculated to maintain the price in the market, but to lull the shareholders into a state of false security, by holding out promise to the ear, have called not only the amount of capital, being 400,000*l.*, or 20*l.* per share on 20,000 shares, but have lately, as appears by the letter of a correspondent, made a further call of 40,000*l.*—the market price of the whole concern being not much above the amount of the last call made.

We trust that we have said enough for a while to awaken the shareholders in these several companies from the stupor which has been productive of so much injury to their own interests, as well as to mining enterprise generally, from the evidence it exhibits of want of care and attention on the part of capitalists, and the consequent result of directors and jobbers making their fortunes, in many instances, at the public cost. The general meeting of the Rhymney Iron Company takes place next week; let some independent proprietor require, on the part of the shareholders, that a reporter of the press shall be present, and we promise him that the proceedings, with some comments, shall appear in our next Number; for ourselves we shall make our ordinary application, and expect, as usual, to be met with a refusal.

We are glad to find that Government is at last directing its attention to the "manufacture" of joint stock companies, and more especially those which come under the designation of *assurance* companies. The injury already sustained by the public from the want of a proper check over establishments of this nature, has been at last too strongly manifested to allow the system so generally pursued to be passed over with impunity. The late exposure of the fraud practised under the title of the West Middlesex Insurance Company, by which 200,000*l.* was abstracted from the industrious portion of the community—the Royal Union Annuity Company, in which Mr. GLENNY cut so conspicuous a figure, and of which the late Duke of YORK and several noblemen were represented to be patrons—the British and Australasian Bank in like manner—the known state of several life insurance and loan companies at the present moment, the investments of which have been either wastefully expended, or advanced without *bond fide* security (the directors trusting to the lives not falling in until they are out)—these have forced themselves on the attention of the Ministers, and it is with satisfaction we learn that they purpose referring the consideration of companies of this nature to a select committee. We need hardly say that the proposed measure has our earnest support. Let the committee be once appointed, and we will undertake to submit to their consideration some questions to be put to the managers and directors, which, if we mistake not, will reduce the number of our assurance companies very considerably, and strike off the list of directors many whose names are a disgrace to those with whom they are associated.

We perceive, by last night's Parliamentary reports, that Mr. LABOUCHERE moved for a select committee to inquire into the state of the law respecting joint-stock companies (banking companies excepted), with a view to the prevention of fraud, which was agreed to.

It is with regret we have this week to record the defalcation of a gentleman, whose name has so oft appeared in our columns as a director of several companies, and who held the responsible office of official assignee in the Court of Bankruptcy—Mr. PETER HARRIS ANNOTT. We felt it our duty, when that gentleman was in the zenith of his glory, living on the property and plunder acquired from others, to expose his conduct, trusting that it would serve as a caution to those who might become connected with him, for we were well assured, from the *exposé* in the matter of the Adelaide Gallery, and other circumstances which had come to our knowledge (of a nature to which we did not feel at liberty to give publicity), that the game must soon be up—and so, unfortunately, it has proved. The amount for which Mr. ANNOTT is in default has been reported at from 60,000*l.* to 80,000*l.*; and we are led to believe that it amounts to nearly, if not full, the former sum. We regretted to find that the *Times* lent itself to glossing over the misconduct of the party, which we can only attribute to the writer being, possibly, an official assignee himself; but the communication made by the LORD CHANCELLOR to the Commissioners of Bankruptcy, of the determination to have monthly returns, will, we have no doubt, put all upon the *qui vive*, and have the effect of securing to the creditors the little which may be left to them from the bankrupt estate. We have heard of one or two bad cases, but as an inquiry is being instituted, we await the result. Mr. PETER HARRIS ANNOTT, and other jobbers like him, will find out that, in the end, "honesty is the best policy."

**T. S. PARSONS.**—No tidings of the President had reached Liverpool up to ten o'clock yesterday (Friday) morning. The pocket-ship *Fingertine*, which would not leave New York earlier than the 14th ult., was then off the port; she may, perhaps, bring some intelligence of the emperor.

Mr. Marchion, the President of the Geological Society, has just left London upon a tour to the Ural Mountains and distant parts of the Russian Empire.



## ON THE NATURE AND PROPERTIES OF SULPHUR.

The activity which at present prevails in the county Wicklow for increasing the supply of the sulphur ores of that district, attracts much attention, and the exposition of any information connected with the nature of the mineral, and the most advantageous manner of employing it in the industrial arts, must also be a subject of great interest to all those who can appreciate the national value of mineral wealth, as well as a great source of riches to those who successfully employ such information. In the first place, a short sketch of the statistics of sulphur, or brimstone, with an account of its nature and combinations, may be useful. It is most abundant in nature, being found in all formations, from the oldest to the newest; but it abounds chiefly in volcanic districts, where it is often found in a native state; it is largely combined with metals in its mineral state, in the form of sulphurets. Copper, iron, antimony, zinc, and lead, are almost always combined with it to a greater or less degree, and it is one of the most important operations in metallurgy to expel sulphur from the ores. Hitherto the sulphur consumed in Great Britain, amounting annually to about 30,000 tons, has been obtained from Sicily, where it is prepared in a manner hereafter described. It is found in mechanical mixture with earth, both at Solfatara, in Sicily and in Poland. Letters have lately been received from Tripoli, giving an account of an attempt made to work the sulphur mines in the Gulf of Syrte, which promise to be very productive. In Sweden, Saxony, and Bohemia, it is found in chemical combination with iron and copper. This mineral is known by the name of "mundic," also "pyrites," from the Greek word, signifying fire, on account of its peculiarity in emitting sparks when struck with any hard substance; it has been used as a substitute for gun flints. The mines of Cornwall, Anglesea, and Wicklow yield this mineral in great abundance, but sulphur has never been extracted from it in this country, except at the Pary's Mine, in Anglesea, where a small quantity is saved, but by a most extravagant process.

It will be interesting to mention a few of the peculiarities of sulphur, as the phenomena it presents, when heated, offer a striking contrast to the received theories of the tempering of steel and glass, and are analogous to those exhibited by Brony. At a temperature of 110 deg., sulphur is very fluid, and of a bright citron colour; it preserves these characters up to 140 deg.; but when once it passes this temperature, the phenomena it presents are most curious; for example, at 160 deg. it commences to thicken, and has a reddish tinge, and, if the heat is continued, it acquires a consistency, such that you may overturn the vessel which contains it without displacing the sulphur; between 220 and 250 deg. this change is most remarkable; towards the boiling point it again becomes liquid, but does not lose the colour given it by the heat, nor does it become so liquid as it was at 109 deg. When raised to a high temperature, and suddenly cooled, sulphur becomes soft; and if the experiment be well conducted on sulphur raised to a temperature of 230 deg., it will become so soft and ductile that it can be drawn into wires several feet in length.

## ON THE USES TO WHICH SULPHUR IS EMPLOYED.

Soft sulphur can be employed with success in the arts, for taking delicate impressions and casts, such as of coins, medals, and seals, or of designs in relief; for in a few days it resumes its premature kindness, and these impressions serve as a matrix for forming other casts. Melted sulphur is also used for these purposes alternately with plaster, as the sulphur contracts in solidifying, while the plaster swells—thus the alterations in the one body are corrected by the other, preserving a proper image, which cannot be done when plaster alone is used. Sulphur is employed to cement iron and stone; it forms sulphurous acid, and by this means sulphuric acid—a commodity we shall treat more fully hereafter; it is used in the manufacture of matches and gunpowder, combined with magnesia, lime, and p-tash; it is used as a medicine, and with mercury it forms cinnabar or vermillion.

## ON THE REFINING OF SULPHUR FROM EARTHY MIXTURES.

When sulphur is found in combination with earthy matter, its purification generally consists of two distillations; the first roughly performed on the spot where it is obtained, with the object of rendering the cost of carriage less expensive; the second is made with more care, near the spot where it is brought to market. At Solfatara the first distillation is executed in a furnace, or gallery, in which are arranged ten or twelve earthen pots, about twenty inches apart, in two ranges—each of these pots containing about thirty pints. When filled with ore, broken to the size of road metal, the top is luted down, but there is a tube connected with an opening in one of the shoulders of the jar, about two inches in diameter and fourteen inches long, which communicates with a second jar, pierced with a hole at the bottom, from which the sulphur flows into a tub of water, and is then condensed—it is sublimed in the first jar and cooled down in the second. At Marsailles there is a large establishment for the refining of sulphur, conducted by M. Michel, who invented the apparatus; it consists of a cast-iron retort, and a vast chamber, which serves as a condenser. The retort, containing about 1500 or 1600 lbs. of material, and is heated by a furnace, which, however, has no communication with the chamber or the retort; an iron door in front serves to clear and recharge the retort, the back of which conducts the fumes of the sulphur to the large chamber, where they are condensed and collected either in a liquid state or as the flowers of sulphur; the temperature of the condenser alone effecting the condition in which it is obtained. Much care is necessary to be observed in this process, as the admixture of a very small portion of air with the fumes in the chamber give rise to explosions, which are sometimes dangerous, for it often happens that the temperature of the chamber is carried to the height of 150 deg. (sufficiently high to inflame the sulphur); sulphuric acid is then rapidly formed, and in this process much heat is given out, and an explosion always follows; by valves placed at proper places this may be avoided, and the chamber should be freed of its oxygen by burning sulphur or charcoal in it before commencing to operate. Of late years the method which has been adopted, termed "decarbonation," has greatly removed the difficulty of the process, as explosions are now seldom known to occur. This arrangement consists in placing a large vessel, filled with crude sulphur, above the retort of distillation, and allowing the heat of the chimney to act upon it; a tube is made to wind round this vessel, furnished with valves, to permit the melted sulphur to flow through it and communicate with the bottom of the retort containing the sulphur to be distilled; this disposition has the advantage of maintaining a more uniform temperature, and rendering the operation more rapid. By the ordinary method the loss amounted to about 18 or 20 per cent., but by M. Michel's modification the loss is reduced to 11 or 12 per cent. Notwithstanding these precautions, the fumes of sulphuric acid which escape when opening the chamber to extract the sulphur or clear it out, frequently destroys all vegetation in the neighbourhood; it is, therefore, important to construct them so that the sulphur can be withdrawn without entering the chamber. Eight or nine charges are usually evaporated before running off the sulphur. It is of great importance to have thermometers inserted in the chamber, at such places as they can be observed, as we have seen that, if the sulphur is above the temperature of 110 deg., and suddenly cooled, it loses its citron colour, and is not so valuable in the market, although equally pure.

## ON THE EXTRACTION OF SULPHUR FROM PYRITES.

Sulphur is also extracted from some metallic sulphurets, and is an important operation in metallurgy; it is more profitably obtained from the persulphuret of iron than from copper ores, and that the process may be more easily be comprehended, it may be said that iron pyrites is combined in such a manner, that if one-half of the sulphur be extracted, the residue will then be constituted in such proportions, that if the sulphur be acidified, and the iron be transformed into the protoxide, then will result the neutral salt of the sulphate of the protoxide of iron; this may be effected by the action of heat; but in Saxony and Bohemia, where sulphur is extracted from pyrites, not more than 25 per cent. of the sulphur contained is obtained, as it is stated that the heat necessary to decompose the remaining sulphur causes the residue to run into a cinder, and that it is impossible to withdraw it without destroying the apparatus. Under slightly different circumstances, however, I have seen the residuum of some sulphur ore from Wicklow, from which 75 per cent. of the sulphur had been extracted at chemical works in Glasgow, which was not so much run as to have rendered it impossible to withdraw it from the apparatus used in the above-mentioned process, and which may be described as follows:—Earthen tubes, open at both ends, and slightly conical, are placed across a furnace or gallery, having each a slight inclination of about an inch; in their lower opening is placed a star of earth, which permits the vapours to escape, but retains the ore, which is broken in small pieces, and

put into the tube; a lid is luted to the upper end, and another tube is attached to the lower extremity, which serves to conduct the sulphur vapours to a condenser containing water; it is then remelted and run into moulds, such as we receive it. Each furnace is provided with from twelve to twenty-four tubes, each containing about one-quarter of a cwt., and the distillation lasts about eight hours; in a furnace of twenty-four tubes about 2000 lbs. of sulphur ore are treated in a week, and about 190 lbs. of sulphur ore obtained, which is about 14 per cent. M. Dattiquet, at his establishment in the neighbourhood of Ramur, has made a slight modification of this apparatus, and obtains the same result. Sulphur is also extracted from pyrites at Falun in Sweden, and Goslar in the Harz Mountains; at the former place it is obtained by a process proposed by the celebrated Ghan, and may thus be described:—On the slope of a hillock a pile of pyrites is laid upon a bed of wood; the wood is first ignited—the pile is then covered with baked earth or puddle (a small opening only being left at the top to regulate the combustion, and which can be closed by a flagstone)—the vapours are allowed to pass along a conduit, about forty-three feet in length, and formed of boards; the vapours which condense in this passage are taken out of small reservoirs in the form of sulphur, and the uncondensed vapours pass into a large chamber, where they circulate till condensed, which can be assisted by artificial means. This chamber, as well as the conduit, should be perfectly airtight, otherwise the fumes of sulphur will be converted into sulphurous acid, and, should moisture be present, into sulphuric acid, by deriving oxygen from the atmosphere; the sulphur in general obtained by this process does contain so much sulphuric acid, that it is necessary to wash it. By the Goslar process not more than 1 or 2 per cent. of sulphur is obtained, but this process is instituted chiefly for the roasting of ore and not for obtaining sulphur; it is also practised at the Pary's Mine, in Anglesea, and may be described as follows:—In the centre of a truncated pyramid a chimney is formed of billets of wood, and, for the purpose of assisting combustion, passages are formed under the pile communicating with this chimney; the mass is then fired by throwing some ignited charcoal down the funnel, when combustion quickly takes place; the pyrites, in decomposing, emits a sufficient heat to evaporate the sulphur, which, at certain stages, is ladled out of small holes made in the top of the pile—the operation lasts for about three months. The sulphur thus obtained in Anglesea contains about 7 per cent. of impurities, while that imported from Solfatara contains only 3 per cent., and that not at all arsenical.

## ON THE NATURE OF THE RESIDUE AFTER EXPELLING PART OF THE SULPHUR.

We have seen that, in Saxony and Bohemia, as well as by the process of M. Dattiquet, that only 25 per cent. of the sulphur present in the ore is extracted, which circumstance would indicate that the residue is a combination of determinate proportions of such a nature, that the persulphuret being formed of two atoms of sulphur and one of iron, this residue will consist of two atoms of iron and three of sulphur; or that it will correspond to the sulphuret of the red oxide of iron, from which can be formed the neutral sulphate of the protoxide of iron; by exposure this salt becomes the sulphate of the protoxide of iron, which, in combination with the necessary quantity of water for crystallisation, forms copperas. Sulphuric acid is the most important of the combinations of sulphur, and a sketch of some of its properties and formation may now be given the more fully, to explain the process by which it is obtained from iron pyrites, which is now so generally coming into use, to the great benefit of so many of our mining districts, but chiefly to the mines of the county Wicklow.

Sulphuric acid is known to us in three distinct forms—first, pure or dry acid; second, combined with water, or what is usually termed, the sulphuric acid of commerce; third, in the state called glacial, or fuming, which is a mixture of pure and hydrate acid in variable proportions. Dry sulphuric acid is solid at the ordinary temperatures—liquefies at 25 deg., and instantly goes off in vapour; when crystallised it is with difficulty redissolved; its crystals are of the form of needles and stars, of a clear transparent white colour; when liquid, its density is 1.97 to 2.00—solid, it would be something more. Ordinary sulphuric acid, or hydrate acid, is an oily liquid, but not so heavy as the anhydrous acid; its density at 15 deg. is only 1.848; neither is it so caustic, although sufficiently so to decompose with rapidity either animal or vegetable material. While the anhydrous acid boils at 25 deg., the hydrous only boils at 310 deg.; the anhydrous congeals at temperatures below 25 deg., while the hydrous only solidifies at 16 or 12 deg. below zero, of the centigrade scale; it does not fume in the air, but it rapidly extracts moisture from the atmosphere, and will imbibe four times its own weight of water if exposed for a sufficient length of time, and frequently agitated. When mixed with water in certain proportions, it gives out various degrees of heat, and great cold is produced by mixing it with ice.

The discovery of sulphuric acid is due to Basile Valentine, a chemist, who lived towards the end of the fifteenth century, but it has since occupied the attention of all who have devoted themselves to chemistry. Mons. C. Desormes has explained the theory of the formation of this acid, as follows:—Having first exhausted a glass globe, he passed into it two measures of sulphurous acid and half a measure of dutoxide of nitrogen (these gases mix without uniting), two measures of oxygen were then introduced into the balloon, when immediately red vapours, due to the formation of nitrous acid, appeared, caused by the combination of the dutoxide with the oxygen added; a few drops of water being then introduced, the red vapours disappeared, and small white crystals began presently to form on the sides of the globe. According to M. Clement Desormes, these crystals are composed of sulphuric acid and dutoxide of nitrogen, united with a certain portion of water. If, at this stage, a greater quantity of water be injected, the crystals will immediately dissolve with a hissing noise, and the temperature will be sensibly raised. The water charges itself with sulphuric acid, and disengages the dutoxide of nitrogen, which, coming in contact with oxygen, returns to the state of nitrous acid, and the red vapours again appear. In this case the water added has determined the separation of the sulphuric acid, and the nitrous acid, which had given one portion of its oxygen for the formation of the crystals, is reduced to the state of the dutoxide of nitrogen, which is disengaged; but this dutoxide of nitrogen again meets with the oxygen and sulphurous acid in the balloon, where again it passes, first into nitrous acid, then into the small crystals before-mentioned; these are in their turn decomposed by the water, and so on until the sulphurous acid or oxygen be entirely consumed. We thus see how a small quantity of nitrous acid can transform an infinite quantity of sulphurous acid into sulphuric acid. M. Gay Lussac has, however, thrown some doubts on the manner in which these elements are said to be in combination; and M. Davos is disposed to think there are some modifications, under certain circumstances, with which we are not fully acquainted.

## ON THE MANUFACTURE OF SULPHURIC ACID.

Sulphuric acid is manufactured on a large scale, by permitting the fumes of sulphur, burned in a separate chamber, or the sulphurous fumes from ignited pyrites, to enter a large leaden chamber, having a few inches of water at the bottom of it, and causing the dutoxide of nitrogen to mix with it there; in some instances, and for the sake of economy, the dutoxide is obtained by decomposing sugar or starch by nitrous acid, thereby obtaining oxalic acid; but, for the most part, the nitrous acid of potash is placed upon shelves amidst the fumes of the sulphur, thereby supplying dutoxide of nitrogen and sulphurous acid in proportions sufficiently economical. The purposes for which sulphuric acid is applied in the arts, are as follows:—In the decomposition of salt for obtaining soda—largely used in making glass, soap, &c., also muriatic acid, which is employed in many purposes; it is the means of obtaining chlorine, and is used in the fabrication of nitric acid, as well as most of the acids known; it separates silver from gold—is an ingredient in the formation of alum—the sulphate of copper and iron; it is partly the means of bleaching, and is used in the preparation of sugar from beetroot; it is used in dyeing, and for a number of other purposes, as a secondary agent. The facility with which sulphuric acid is manufactured from pyrites, renders the distillation of sulphur a matter of less importance than would at first appear; yet the immense expense of carriage of so much extraneous matter is a subject well worthy of consideration; and having given the analysis of the nature of the residue, after expelling different portions of sulphur, we may notice a few of the useful purposes for which that residue may be employed.

## REMARKS ON THE EFFECT EMPLOYMENT OF TWO DEGREE OF DISTILLATION.

We have already seen that when 50 per cent. of the sulphur contained in the persulphuret of iron is expelled, the sulphate of iron, or copperas, may be obtained from what remains. We are, however, told that not

more than 25 per cent. of the sulphur of the pyrites can be obtained without injury to the apparatus, and that, in this case, the residue would contain that portion of sulphur corresponding to the sulphate of the peroxide of iron. Now, either of these can be turned to a profitable account. Copperas is an article of commerce in large demand; and both the sulphate of the protoxide and the sulphate of the peroxide are the basis of the fuming sulphuric acid of Nordhausen, which, although not of such general use as the ordinary acid, is yet essential to dyers for forming solutions of some of their dyes, for which the ordinary acid will not answer. At present there is but one manufactory for the formation of this acid; the internal economy of this establishment is kept secret, but we are well acquainted with the manner of obtaining this acid; the residue in this case is the red oxide of iron, long known by the name of "cokeothar," or "crocus of Mars"—a powder used for giving a fine polish to steel; also used by silversmiths under the name of "rouge," and by opticians for polishing the specula of telescopes.

By a systematic arrangement of this kind, every particle of matter in the pyrites may be turned to account, and on the spot where the pyrites is obtained a very perfect process might be established to supply the market with pure sulphur, the value of the residue being sufficient to defray the cost of its extraction—time being the chief agent for converting the residue into the marketable commodities mentioned. The expense of fuel requisite for the distillation of the sulphur may be avoided, by applying to that purpose the heat given out by the combustion of pyrites in the act of forming sulphuric acid, and a great portion of the ore, after having given off one part of its sulphur by distillation, may then be used for the manufacture of sulphuric acid, where it will be still further reduced. It is quite possible to expel all the sulphur from the iron by roasting, but it involves considerable expense in fuel; at the present time an experiment, on a large scale, is being made, to test the economy of the operation. The formation of sulphuric acid of commerce, at the same period in the fabrication of the fuming acid, would be economical, as the heat given out would distil the dry acid from the salt of the sulphate of the peroxide of iron, at the same moment as it was driving off the sulphur from the pyrites, in order to form both sulphur and the base of the salt mentioned.

## ORIGINAL CORRESPONDENCE.

## APPLICATION OF DRY AND WET COPPER ASSAYING.

## TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Without waiting longer for the verification of "Observer's" implications, I may just suggest to him, that whilst reasoning and arguments bear their own weight, and are sometimes best anonymous, to keep clear of personality—opinions and assertions, on the contrary, rest upon authority, and it is hardly honest to give them the weight of names who have not been consulted. The following observations on the respective applications of dry and wet assaying I should have supposed needless, but for the misunderstandings, on this point, which have appeared among your correspondents.

There are two objects in assaying—one to ascertain the entire proportion of metal present—the other, to learn the quantity which can be profitably extracted. The first, for the satisfaction of the smelter, showing how much is lost in his operations, and stimulating to their improvement—the second, to ascertain the market value of the ore between buyer and seller. The first object can be effected only by the wet process, of which more presently; but it appears to me that the miner will only deceive himself if he trusts to this process to ascertain the value of his ore. Their relative value depends, not on the proportions of metal contained, but on the quantities which can be extracted with profit; and it would be a sad mistake to reckon an ore which gave, by the wet process, 2 per cent. of copper, at 1.20th the price of one giving 40 per cent.—to take the first at as many shillings per ton as the other pounds. Even by the dry process this will not always be the case; but, by making our assay as nearly as possible a miniature imitation of the smelting process, we learn, not what the ore contains, but what the smelter can get from it—and, allowing for his expenses and profit, can thus compute the value of the sample. Nor is this the only advantage of dry assaying—it returns the metal in its marketable state, and by operations at once distinct and expeditious. For commercial purposes, therefore, the dry assay is the most trustworthy—and, for distinctness and expedition, the Cornish method is excelled by none, whilst, in good hands, it may be equally relied on for comparative accuracy. How far this comparative accuracy is all the smelter has to expect, is another question—Whether improvement in assaying ought not to keep pace with those in smelting, and whether buyer and seller would not be likely to agree better, if the mystery of "surplus copper" was fully laid open? There is nothing impracticable in this; nor need assayers be told that ores can be worked cleaner in the crucible, where the cost of fluxes is but an object, than in the great furnace, where economy is essential. But we will not anticipate Mr. Thomas's promised lecture.

Wet assaying, to the miner, is chiefly an object of curiosity, but may also serve as an occasional comparative check upon the dry produce—its general utility is, unquestionably, to the smelter. A new kind of wet assay has been lately introduced, and has occupied a place in your columns—the electro-chemical, or voltaic. This is very effective, and where porous earthenware vessels are not easily attained (for I cannot recommend the Cornish crucible for this purpose), pig's gut, or even sheep gut, answers very well. The objection of the length of time required is not very serious, as, if we put it in action at night, it is done in the morning; but the exact precipitation of the copper, where an ore has been dissolved in muriatic acid by aid of nitric, requires some practice—perhaps as much as to attain correct results from the common wet assay. Of this process (the common wet assay), the fault is not that it gives too little copper, but that, without especial precautions, it gives too much. These precautions are given in the *Mining Review*, No. XIX, July 31, 1839, and for those who have a sand heat fit to work with sulphuric acid, the following directions may be unnecessary; but as this is not possessed by many of those interested in copper assaying, some other practical instructions may be not unacceptable.

The ore, in fine powder—say 100 grs.—is to be heated in muriatic acid, and nitric acid to be added, in small portions, at intervals, shaking all well together at each addition, until the last portion of nitric acid is not decomposed, and all the metallic matter is dissolved. The proportions of both acids may be varied, according to the quantities of metal present; 1 oz. of muriatic acid is generally enough for 100 grs., but 14 oz. may be used if the ore is very rich, and a drachm of nitric acid will generally be enough for the richest ore—it may be added ten drops at a time, at first pretty quickly—say at intervals of two or three minutes; but as the effervescence and red vapours become less copious, the intervals must be longer—say five to ten minutes. As the metallic part falls first, after shaking, a little practice enables us to distinguish when some of it is left undissolved, which may easily be effected within an hour. It is desirable to use as little nitric acid as possible, since any excess of this acid is unfavourable to the exact precipitation of the copper; on the other hand, a sufficient proportion must be employed, else, whilst most of the iron present will be dissolved by the muriatic acid, copper will certainly be left in the residue. The solution may be evaporated to dryness on the residue, or first poured off and evaporated separately, washing the residue with water, and adding this water to the solution for evaporation. I prefer the latter method. We may also, now, convert the solution into sulphate, by addition of sulphuric acid during the evaporation. This is attended with some advantages, as getting rid of any residue of nitric acid, and of any lead which the sample might contain. It is not generally necessary where excess of nitric acid has not been used, but may be considered generally advantageous. The proportion of sulphuric acid is not required to be very exact. It may pretty generally be taken at a minimum measure for each grain dissolved. For this purpose, the solution being poured off, the residue washed with a little water, and the washings added to the solution, the residue is to be dried and weighed. The difference between its weight and the 100 grs. employed gives the quantity dissolved; for example, suppose the residue 65 grs., the quantity dissolved is 100—65=35 grs.; add these to the solution, thirty-five minims (drop measures) of sulphuric acid, and proceed with the evaporation. The acid must be added cautiously, if the solution is hot. It will drive off the muriatic and nitric acids, as well as the fluxes, if flux is present, and leaves the dry mass in the state of sulphate. This may now be treated as directed in the *Mining Review*, above quoted, except that, as the heat employed is not supposed great enough to boil away the sulphuric acid, the further addition of this acid will not be requisite. I have been in the



practice of using only spirit of water for dilution, having never found any advantage in employing 2 lbs., as there ordered. The carbonaceous matter is easily washed away, as there stated, but this is not so fully the case with arsenic of iron, which sometimes falls with the copper when muriatic and nitric acids are the solvents—nor is it easy to prevent oxidation in drying. Both these cases falsify the assay by increasing the apparent produce in copper, and, in fact, it generally gives too much. It may be corrected by melting the produce with black flux in a small crucible, or even in a tobacco pipe, stopped at bottom, but much more readily by Plattner's process with the blowpipe, which, as this letter is long enough, shall be described in a further communication. Your's, &c.,  
Plymouth, April 1. J. FRAIDRAUX.

#### GENERAL MINING ASSOCIATION.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—No report of the proceedings of this company having ever appeared before the public, I think some notice of their late unwarrantable conduct towards the shareholders at their last meeting, held on the 11th of February, should be made known, I therefore, hope you will give this statement a place in your Journal.

This company was formed by Messrs. Randel and Co., in 1824. After holding out the most brilliant prospects for seventeen years, the full amount of shares were paid up last year. For the last three years so great were the profits represented to be, that assurances were made to the shareholders that, instead of any further deposits being called for (17l. a share at that time only having been paid), the following year a dividend, the shareholders might be assured, would be announced; when the time arrived some plausible pretext was brought forward for a further delay of the promised dividend, and the unexpected necessity to make a further call of 11s. a share; and thus have the shareholders, for the last three years, been duped, until the full amount of 20l. per share was last year paid up. At the above-mentioned meeting the directors (who, I have strong grounds for believing, have themselves been benefitting) concoct a scheme to extort a further sum of 2l. a share from the unfortunate shareholders, offering to such as might consent to such a further outlay 6 per cent. interest, to be paid out of the first profits, should any ever arise out of this unfortunate speculation; and the directors have the impudence to exclude all shareholders who will not submit to this further demand from any profits which may arise out of this concern till the 6 per cent. is paid—thus, in a clandestine manner, reaping any advantages amongst themselves. In all other companies a certain number of the directors go out annually, and their reports are made openly to the public, but this company, like the mysterious proceedings of the Inquisition, sit in secret council to enrich themselves out of the purses of the unsuspecting shareholders.

I am, Sir, your obedient servant,

London, March 26.

AN ORIGINAL SHAREHOLDER.

[Such proceedings must be expected from all companies who dare not face the light. Why do not the shareholders, at the general meetings, insist on the reports for the public press being present. We remember well the game played on the formation of the company. We then made extracts from the correspondence, for insertion in the Mining Review, but when it was found that we would not take garbled extracts, but preferred copying from the original correspondence for ourselves, we were no longer permitted to have access to the books of the company. We repeat, that it is the fault of the shareholders, they do not enforce those rights to which they are lay claim. Who, we would ask, are directors? but the paid officers of the company. In them is vested the patronage of appointments—they receive their salaries, undertaking to protect the interests of the shareholders, and to see that the sub-agents do their work—acquire information, in their public character, of which they oft avail themselves for private interest—and yet are they regardless of public opinion, and, in too many instances, as a body, pursue a course which, as individuals, they would be ashamed to acknowledge. The late failure of a director of numerous companies, whose character as a merchant stood pre-eminently high, and who, moreover, filled the office of Bank director, should be a lesson to the many.]

#### BRITISH IRON COMPANY.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—At the special general meeting of the British Iron Company, on the 25th ult., Mr. Larpent (the chairman) uttered aspersions on my character which I consider too serious to pass unnoticed. I therefore addressed a letter to him on the subject, on Tuesday last, but, as he has not thought proper to give any reply to that letter, I rely on your justice to insert the enclosed copy of it in your Journal to-morrow, which has already, with so much ability, attracted public attention to the concerns of this ruinously-conducted company. I am, Sir, your's, &c.,  
RICHARD CURT.

[Mr. Curt will, we feel assured, appreciate the motive which influences us in declining to give insertion to the copy of letter addressed to Mr. Larpent, accompanying the above. We regret that many questions of an important nature are too frequently lost sight of by personalities being indulged in; and, in the present instance, we feel that the real question at issue may be merged or forgotten in the consideration of attack and defence of personal character. We have too high an opinion of Mr. Larpent, and the position he holds in society, to suppose that he would avoid entering into the general charge, and merely amuse or divert his auditory by a personal attack on Mr. Curt (which is the ground Mr. C. assumes); and, on the other hand, we should hope that the latter gentleman, having taken upon himself a Herculean task—that of cleansing the Augean stable—according to his own representations, will not allow himself to have his attention drawn from the subject. We have perused the letter addressed to Mr. Larpent, and, so far as Mr. Curt's statements go, we must say that the attacks, which we are given to understand were made, reflect discredit upon the author; but anxious as we are to expose abuses, and at all times to go with the shareholders in an inquiry, we must be excused if we decline giving insertion to a letter which, however proper it may be considered as between individuals, is not, in our opinion, one fitted for a public Journal. The letter, we think, might have been written in a more cool and temperate manner, for it appears to us that the style itself precludes the chance of a reply.]

#### ON DIALLING.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—You have kindly permitted the insertion of many letters on dialling in your much-valued Journal, since the subject has been exhausted, in the hope, I suppose, that some practical information might be extracted from them. Presuming on your indulgence, I beg to offer a query or two, put in the form used by Mr. Budge, in your last paper, but the answers differ materially—

Query.—Suppose a subterranean survey has been made, containing various details, it is practicable to throw all these details into one line, and ascertain the exact length and direction of that line?—Ans. It is practicable, but by no means necessary; nor is it free from doubt and uncertainty as to that method which has been called "the descriptive tracing and pegging method," and for this reason—there is a greater variety of operations of the mind in the use of tables, and every question of the mind is subject to error. Wrong numbers may be used when the work is performed by the use of logarithms, but if any geometrical construction be employed, great danger of error arises from the reduction of the scale. There is no reduction of the scale in the common mode of "tracing and pegging;" nor any danger from various operations of the mind—all is straight forward—and it requires no more than a "count—some table"—error.

Q. But how do you prove your work?—A. I cover the ground a second time, both underground and at surface, making the second series of shafts underground longer or shorter at pleasure. Thus the second series, on the surface, if the last peg of the first series answers to that of the second series, the work is right. No mathematical calculation can give stronger proof of correctness than this; nor is there any easier, cheaper, or more correct method of taking a survey.

Mr. Budge observes, in conclusion—"That our opponents have not ventured to touch the Gwennap case, even with the point of a feather."—No, indeed, but the new version of the Gwennap story is evidently scratched into his memory with the hard point of a diamond.

I am, Sir, your obedient servant,

Perrow, March 30.

R. THOMSON.

#### COLLIERY INSTRUCTION.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—The annexed paragraph is taken from a northern paper—"The death of a series of lectures, colloquial and scientific subjects, that have been delivered once a fortnight to the inhabitants of Allotment, was given at that place on Monday evening, by Mr. Hunter, viewer, Backworth Colliery, to a numerous and attentive audience. Among other interesting subjects brought before the meeting, there was an account of the coal strata in this district, and of the 'fractures' which have been encountered at various collieries; and these were well illustrated by diagrams and sections executed in an excellent manner. The thanks of the meeting were cordially awarded to Mr. Hunter, and as it must be admitted that the best consequences would result from imparting useful information to the colliery population, some means might surely be adopted at every colliery for presenting the mental improvement of pitmen."—*Newcastle Journal*, 26th March, 1841.—I send it to you, in hopes that, if convenient to insert it in your widely-circulated Journal, it will get more publicity amongst miners,

and may induce others to follow the laudable example of Mr. Hunter, "in imparting useful information to the colliery population." The improvement of the mental, moral, and social condition of the people must ever be a subject of deep interest with the wise and good; but while endowed universities, colleges, and schools have been provided for the rich, little has been done in this way for the poor. This, probably, may be partly accounted for by that most absurd feeling of degradation usually attached to personal labour, and particularly the labour of the miner. Those who move in what is called fashionable life, erroneously imagine that the minds of the working classes are as coarse as their manners, and that to attempt to cultivate either is Utopian, or at least a Herculean task; this, however, is not the fact, we have every day evidences to the contrary presented to our notice. What working men lack in intellectual instruction, it too often comes to them through the sorrows of the heart—witness the philanthropic spark that is lighted up in the breast of the poor miner when any of his fellow-workmen are in danger, as he rushes to the rescue, and not unfrequently falls a victim to his own temerity; on other occasions he contributes out of his slender income to the necessities of his comrades when sickness or other misfortunes assail them. Many other good qualities of the heart might be recounted, which ought to put to the blush those who, "arrayed in a little brief authority," treat the poor man with insolence and as an inferior being. Working men are not only greatly interested in the advancement of education amongst those of their own class, but also in getting a better education for those that are set over them.

March 29.

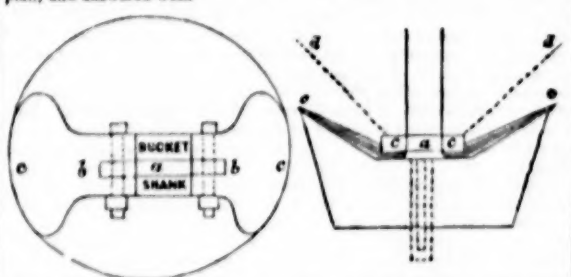
A WORKMAN.

[We are obliged to our correspondent for having directed our attention to the series of lectures referred to in his letter. It is highly praiseworthy on the part of Mr. Hunter (as a colliery viewer) to afford to those under him, as well as others employed in collieries, that information which his experience has enabled him to acquire, for it is equally to the interest of the capitalist as it is to the security and well doing of the miner, that the latter should possess more information than simply the use of the pick. Taking agents "from the ranks," is a course we have ever approved, and, without instruction be afforded them, to assist their observations underground, we fear that as is too frequently the case, agents will continue to be appointed who lack that information which can alone be useful, however scientific may be their attainments—that of a knowledge of underground works, the nature of the strata, and the work a man can do.]

#### IMPROVED PLAN FOR RAISING MINE PUMPS.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—I beg to hand you a plan, much simpler than the one inserted in your Journal of the 20th March, for drawing a set of pumps. Thirty fathoms were drawn north of the Tyne lately, sixteen inches in diameter, by it; the whole was done at a tenth of the expense of Arthur and Eddy's plan, and answered well.



The above is done by putting a hole through the bucket shank a, and in it inserting a piece of square iron, b, having two holes—a pair of iron falls, rather long, for the bucket shell, are by a bolt attached to the square iron bar; these are of three-eighths boiler plate, laid with steel at c, and ground; as the bucket is lowered, the falls rise in the direction of d, and find their place, and hold on applying the strain.

I am, Sir, your obedient servant,

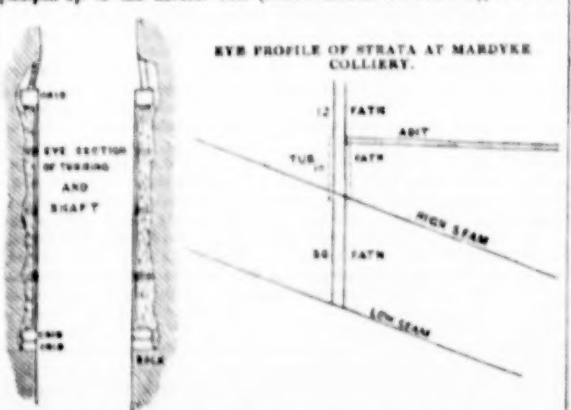
Sunderland, March 23.

EDWARD STANLEY.

#### CAST-IRON TUBING.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Having lately had the honour of directing the application of a cast-iron tub, for the stopping back of water at Mardye Colliery, the property of the Irish Mining Company, and, as it is the first attempt of the sort in Ireland, I submit a short description of it to your notice. The colliery contains two principal seams of coal, lying at an angle of one in three. The upper one, lying at the depth of twenty-two fathoms, is exhausted; and in order to win the second seam, at the depth of thirty fathoms further, the waters of the upper seam were required to be either pumped up to the natural adit (twelve fathoms from surface), or to be



forced up to that point of discharge by tubing. In order to give this project a fair chance, a piece of fire-clay, lying below the first seam, was taken advantage of as a foundation, and the shaft was rounded out to ten feet diameter. The base of tubing is made to rest upon a pile of oak cribs, fitted closely to the fire-clay foundation, and wedged from behind as long as ever a wooden wedge can be driven. This done, the cast-iron tub begins to be built, consisting of cast-iron segments, four feet long, two feet high, and three-quarters of an inch thick, with a rectangular flange all round, of three inches, between each of these segments are placed half-inch (and ways) fir deal, wherein to wedge; the space between the segments and the rock is also stuffed with small stones, and tightened with wood. The top of the segments was completed by a wooden crib, which was stayed fast against the superincumbent rock, and then the whole fabric underwent the most severe wedging so long as any leak continued; and, when finished, the shaft was laid perfectly dry, with the feeder of water discharging out at adit twelve fathoms above, and the sinking of the shaft resumed perfectly dry. The pressure against every square inch of the lower range of tubing is equal to two and a half atmospheres, or about 37 lbs. per inch, and, taking the average altitude at thirty-six feet, the whole tub is sustaining a pressure of about 81,200 tons; and so complete is the job, that the sinking has been since carried on without any pumping apparatus, whilst sufficient water is discharging at the adit as would give employment to a heavy engine.

It is often found convenient to surround these tubs with a sufficient quantity of stone walling, to enable the wedging to be made effective.

Some years ago I effected the "winning" of a shaft, thirty fathoms deep, at Castle Comer, in the same county, by means of a plank tubing, of ten fathoms in length, constructed of three-inch planks, a spiked against wooden cribs, b, and supported again by a range of inside cribs, c, which were in their turn clad with common deals; this mode of stopping water was practised for many years in some, previous to the invention of cast-iron tubing.

I am, Sir, your's, &c.,

Newcastle-on-Tyne, March 22.

M. DEXA.

#### ON THE QUESTION OF POOR RATES CHARGEABLE ON MINES IN IRELAND.

TO THE EDITOR OF THE MINING JOURNAL.

SIR,—Noting in one of your late Numbers some remarks on the question of Poor Rates chargeable on mines in Ireland, I beg to send you a copy of two enactments of the Act of Parliament, from the first of which you will perceive that half of the rate is to be borne by the landlord, and from the second it will be noticed, that any reservation in a lease by the landlord, for freedom from rates, is not applicable to the novel question of Irish Poor Rates. I am, Sir, your's, &c.,  
HIBERNIAN.

Rathfriland, March 29.

74.—And be it enacted, that where the person occupying such property shall be liable to pay a rent in respect of the same, he may deduct from such rent for each pound of the rent which he shall be liable to pay, one half of the sum which he shall have paid as rate in respect of each pound of the net annual value (whether such rent shall be greater or less than such net annual value), and so in proportion for any less sum than a pound.

77.—Provided also, and be it enacted, that any covenant or agreement whereby any person liable to pay rent, and entitled, under the provisions of this Act, to deduct therefrom any rate, or portion of rate, shall have covenanted or agreed, or shall hereafter covenant or agree, to forego such deduction, shall, so far as such rate is concerned, be of no effect.

\* Properties subject to the rates—see Irish Poor Law Act—lands, buildings, open mines, &c.

[We are obliged to "Hibernian" for directing attention to the two clauses in the Irish Poor Law Act, appended to his note. It certainly does appear by clause 74, that the mine adventurer is entitled to deduct one-half of the rate from the rent payable, while clause 77 enacts, that if the tenant has covenanted to pay any rate, or portion of rate, he shall not be called upon to do so, so far as the poor-rate is concerned, and hence the rate must fall upon the landlord. This seems to us very *Hibernian*, and perhaps our correspondent will explain his letter in a postscript next week. Query.—Does rent and royalty bear the same construction.]

#### ON THE PURIFICATION OF GOLD.

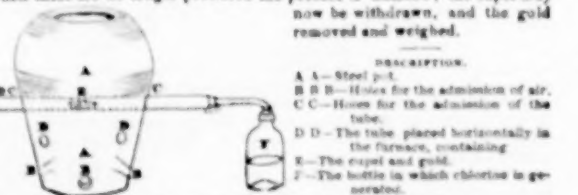
BY LEWIS THOMPSON, ESQ.

[The sum of Twenty Guineas was presented by the Society of Arts to the author, for the following communication on his method of purifying gold.]

In the common mode of assaying gold, the alloy to be assayed is subjected to two operations, cupellation and parting, each of which requires great care and skill; so much so indeed that success seems rather to be the effect of a particular tact on the part of the assayer than the result of a well-defined chemical process. The plan which I now propose for assaying and purifying gold is no less simple in execution than certain in effect, and is founded upon a circumstance long known to chemists—viz., that not only has gold no affinity for chlorine at a red heat, but that it actually parts with it at that temperature, although previously combined; that is to say, the chloride of gold is reduced to the metallic state by heat alone, it cannot, therefore, possess any affinity for chlorine when red-hot; this, however, is not the case with those metals with which gold is usually alloyed, it offers, therefore, at once an easy and certain means of separation. The application of these facts is all, therefore, to which I can lay claim, as the facts themselves have been known for many years, and the reason why they have not been so applied is, that hitherto chemists have not directed their attention to this art, but have left it entirely in the hands of the assayers, who are, for the most part, ignorant of chemistry. The process here proposed has been abundantly tested by myself and others, and employed by those wholly unacquainted with chemistry, as well as by men of eminence in that science, with equal success. There is, indeed, but one source of failure, and this arises from the intense action of chlorine upon the baser metals when melted, by which portions of the alloy are spirited up or projected from the cupel, as happens in the common mode of assaying silver when the heat is too great. This inconvenience is to be avoided in two ways. Firstly, by allowing the chlorine to be evolved slowly at the commencement of the operation, by which the intensity of the action at first is diminished, until the relative proportion of gold in the alloy is increased; or, secondly, by passing the chlorine over the alloy in powder, or laminated into a thin plate at a dull red heat for a few minutes, and then raising the temperature so as to melt it when the fumes of the metallic chlorides have visibly diminished. In conclusion, I can only add, that a very little practice will enable any one in possession of a good balance to make assays of gold with the greatest accuracy. In a course of experiments, conducted at Guy's Hospital, in the presence of Mr. A. Aikin and other scientific gentlemen, a piece of gold was twice alloyed, and then purified by chlorine, without any sensible loss when weighed in a balance which readily turned with the one-hundredth of a grain.

The furnace which I employ for the process is made out of one of those pots employed for melting steel, and which cost about 1s. 6d. each. They are from fourteen to sixteen inches in height, and consist principally of Stourbridge clay and coke. Their form is rather peculiar, as the upper part is contracted so as to form a kind of dome, as in the figure. They are so soft as to be easily cut with a knife; and I have been thus far particular in describing them, because the practical chemist will find them of great use in the laboratory for small furnace operations. One of these pots, then, is pierced near the bottom with four holes, at equal distances from each other and from the bottom; parallel to and between them, but about two inches higher up, another row of similar holes is placed, the whole of which holes should be from a half to three-fourths of an inch in diameter; about three inches above these the sides of the pot are perforated with two larger holes of at least one inch in diameter. These must be diametrically opposed to each other, and upon the same level, i. e. at equal distances from the bottom. The furnace is now finished.

To assay gold, place an earthenware tube in the two upper holes, and light the furnace (a mixture of coke and charcoal answers best, though coke alone will do); when the tube is seen to be white-hot, place in it the alloy contained in a little cupel made of bone-ash, and push it along to the centre of the furnace by means of a wire, then connect one end of the tube with a bottle in which chlorine is forming from a mixture of peroxide of manganese and muriatic acid; the chlorine will, consequently, pass along the heated tube and over the melted alloy, with the silver, copper, &c., of which it will combine and leave the gold pure and untouched. During the process dense fumes may be observed to fill the tube, and when these are no longer produced the process is finished; the cupel may now be withdrawn, and the gold removed and weighed.



#### REPORT OF ARTHUR AIKIN, ESQ., F.R.S., ETC.

The experiments above alluded to, as having been made in my laboratory, were conducted by Mr. Thompson himself, under my inspection. The gold was obtained from an assayer, and was stated to be perfectly pure; but in many instances, on being subjected at a melting heat to the action of chlorine gas, a very small diminution of weight was observed, occasioned, no doubt, by the volatilisation of a little alloy, for the bottom of gold underwent no further diminution whatever on a repetition of the process. The gold thus purified was mixed with silver and copper, or with silver and brass; and, being put into a small porcelain tray, with a little chalk or common salt, was suddenly introduced to the hottest part of the tube. When the alloy was judged to be melted, chlorine gas was passed in at one end of the tube, the other being left quite open or communicating with a small glass retort to collect the volatile products. A dense yellowish vapour almost immediately filled the tube, part of which condensed in glassaceous crystals in the end of the tube; the remainder passed into the retort, lining it with a brownish-yellow crust, or, if a little water had been put into the retort, producing a greenish liquor, which, by the usual tests, was shown to contain chlorides of copper, zinc, and iron. The latter was, no doubt, derived from the ferruginous clay of which the tube was made, for the inside of it, after the process, was found to be nearly white.

On examining the contents of the tray, after the production of vapour had ceased, the bottom of gold was found included in a melted mass of chlorides of sodium (or chloride of calcium, if chalk had been put into the tray) mixed with chloride of silver, the presence of alkaline chloride seeming to have the property of preventing the volatilisation of chloride of silver.

In all the first trials, the bottom of gold was found to weigh considerably less than before the process, and the accidental breaking of one of



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